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Process for the preparation of pesticides

The invention relates to a process for the preparation of compounds of the formula

$$R_9O$$
 X
 Y
 Y
 R_3
 R_4
 $(R_5)_n$
 R_7
 $(I)_1$

and, where appropriate, their tautomers, in each case in the free form or salt form, in which either

X is CH or N, Y is OR1 and Z is O,

or

X is N, Y is NHR₈ and Z is O, S or S(=0);

R₁ is C₁-C₄alkyl;

R₂ is H, C₁-C₄alkyl, halogeno-C₁-C₄alkyl, C₃-C₆cycloalkyl or C₁-C₄alkoxymethyl;

R₃ and R₄ independently of one another are H, C₁-C₄alkyl, C₁-C₄alkoxy, OH, CN, NO₂, a (C₁-C₄alkyl)₃-Si group, where the alkyl groups can be identical or different, halogen, (C₁-C₄alkyl)S(=O)_m, (halogeno-C₁-C₄alkyl)S(=O)_m, halogeno-C₁-C₄alkyl or halogeno-C₁-C₄alkoxy;

R₅ is C₁-C₆alkyl, halogeno-C₁-C₆alkyl, C₁-C₆alkoxy, halogeno-C₁-C₆alkoxy, C₁-C₆-alkylthio, halogeno-C₁-C₆alkylthio, C₁-C₆alkylsulfinyl, halogeno-C₁-C₆-alkylsulfinyl, C₁-C₆alkylsulfinyl, halogeno-C₁-C₆alkylsulfinyl, C₁-C₆alkylsulfinyl, C₁-C₆alkyl, halogeno-C₁-C₆alkyl, halogeno-C₁-C₆alkyl, halogeno-C₁-C₆alkyl, halogeno-C₁-C₆alkyl, halogeno-C₁-C₆alkyl, halogeno-C₁-C₆alkylsulfinyl-C₁-C₆alkyl, C₁-C₆-alkylsulfonyl-C₁-C₆alkyl, halogeno-C₁-C₆-alkylsulfonyl-C₁-C₆alkyl, C₁-C₆-alkylcarbonyl, halogeno-C₁-C₆-alkylcarbonyl, C₁-C₆-alkoxycarbonyl, halogeno-C₁-C₆-alkylcarbonyl, C₁-C₆-alkyl-aminocarbonyl, C₁-C₄-alkoxyiminomethyl; di(C₁-C₆alkyl)-aminocarbonyl, where the alkyl groups can be identical or different; C₁-C₆-alkyl-amino, di(C₁-C₆alkyl)-amino, where the alkyl groups can be identical or different; halogen, NO₂, CN, SF₅, thioamido, thiocyanatomethyl; an unsubstituted or mono- to tetrasubstituted C₁-C₄alkylenedioxy group, where the substituents are selected from the

- group consisting of C_1 - C_4 alkyl and halogen; or QR₆, where, if n is greater than 1, the radicals R₅ can be identical or different;
- R₆ is C₂-C₆alkenyl or C₂-C₆ alkynyl which are unsubstituted or substituted by 1 to 3 halogen atoms; (C₁-C₄alkyl)₃Si, where the alkyl groups can be identical or different; CN; or an unsubstituted or mono- to pentasubstituted C₃-C₆cycloalkyl, aryl or heterocyclyl group, where the substituents are selected from the group consisting of halogen, C₁-C₆alkyl, halogeno-C₁-C₆alkyl, C₁-C₆alkoxy, halogeno-C₁-C₆alkoxy, phenoxy, naphthoxy and CN;
- A either is a direct bond, C_1 - C_{10} alkylene, -C(=O)-,-C(=S)- or halogeno- C_1 - C_{10} alkylene and R_7 is a radical R_{10} ,
 - or is C_1 - C_{10} alkylene, -C(=O)-,-C(=S)- or halogeno- C_1 - C_{10} alkylene and R_7 is OR_{10} , $N(R_{10})_2$, where the radicals R_{10} can be identical or different, or - $S(=O)_qR_{10}$;
- R₈ is H or C₁-C₄alkyl;
- R₉ is methyl, fluoromethyl or difluoromethyl;
- R₁₀ is H; an unsubstituted or substituted C₁-C₆alkyl, C₂-C₆alkenyl or C₂-C₆alkynyl group, where the substituents are selected from the group consisting of halogen; (C₁-C₄alkyl)₃Si, where the alkyl groups can be identical or different; C₃-C₆cyclo-alkyl, which is unsubstituted or substituted by halogen; C₁-C₆alkoxycarbonyl, which is unsubstituted or substituted by halogen; unsubstituted or substituted aryl, where the substituents are selected from the group consisting of halogen, halogeno-C₁-C₄alkyl and CN; a (C₁-C₄alkyl)₃Si group, where the alkyl groups can be identical or different; C₃-C₆cycloalkyl, which is unsubstituted or substituted by halogen; C₁-C₆alkoxycarbonyl which is unsubstituted or substituted by halogen; or an unsubstituted or substituted aryl or heterocyclyl group, where the substituents are selected from the group consisting of halogen and halogeno-C₁-C₄alkyl;
- Q is a direct bond, C_1 - C_8 alkylene, C_2 - C_6 alkenylene, C_2 - C_6 alkynylene, O, O(C_1 - C_6 alkylene), (C_1 - C_6 alkylene)O, S(=O)_p, S(=O)_p(C_1 - C_6 alkylene) or (C_1 - C_6 alkylene)S(=O)_p;
- m is 0, 1 or 2;
- n is 0, 1, 2, 3, 4 or 5;
- p is 0, 1 or 2; and
- q is 0, 1 or 2,

and the C=N double bond marked with E has the E configuration,

which comprises

a1) reacting either a compound of the formula

in which A, R_2 , R_5 , R_7 and n are as defined for formula I and the C=N double bond marked with E has the E configuration, or a tautomer thereof, in each case in the free form or in salt form, if appropriate in the presence of a base, with a compound of the formula

$$R_3$$
 X Y X_1 (III),

which is known or can be prepared by methods known per se and in which X, Y, Z, R_3 , R_4 and R_6 are as defined for formula I and X_1 is a leaving group, or a tautomer thereof, in each case in the free from or in salt form, or a2) reacting a compound of the formula

$$R_{2} = (IV)_{0}$$

in which A, R_2 , R_5 , R_7 and n are as defined for formula I and the C=N double bond marked with E has the E configuration, or a tautomer thereof, in each case in the free form or in the salt form, if appropriate in the presence of a base, with a compound of the formula

$$R_9OX$$
 Y ONH₃CI (V),

which is known or can be prepared by methods known per se and in which X, Y, Z, R_3 , R_4 and R_9 are as defined for formula 1, or a tautomer thereof, in each case in the free form or in salt form, or b1) reacting a compound of the formula

in which R_2 , R_5 and n are as defined for formula I and the C=N double bond marked with E has the E configuration, or a tautomer thereof, in each case in the free form or in salt form, if appropriate in the presence of a base, with a compound of the formula

which is known or can be prepared by methods known per se and in which A and R_7 are as defined for formula I and X_2 is a leaving group, and either further reacting the compound thus obtainable, of the formula IV, for example according to method a2), or

b2) reacting it with hydroxylamine or a salt thereof, if appropriate in the presence of a base or acid catalyst, and further reacting the compound thus obtainable, of the formula II, for example according to method a1), or

c) reacting a compound of the formula

$$O_{R_2}$$
 $(R_s)_n$ $(VIII)_n$

which is known or can be prepared by methods known per se and

in which R_2 , R_3 and n are as defined for formula I, or a tautomer thereof, in each case in the free form or in salt form, if appropriate in the presence of a base, with a C_1 - C_6 alkyl nitrite, and further reacting the compound thus obtainable, of the formula VI, for example according to method b),

the E isomers of the compounds of the formulae II, IV and VI, or a tautomer thereof, in each case in the free form or in salt form, a process for their preparation and their use for the preparation of compounds of the formula I.

The compounds of the formula I are known pesticides. The processes known to date for their preparation give mixtures of E and Z isomers in respect of the C=N double bond marked with E in formula I of different composition, depending on the process. Since the biological properties of the E isomers are in each case found to be superior to those of the mixtures and of the Z isomers, there is a need to develop preparation processes for compounds of the formula I having the isomerically pure E configuration. This object is achieved by the preparation process according to the invention.

Unless defined differently, the general terms used above and below are defined as follows.

Carbon-containing groups and compounds in each case contain 1 up to and including 8, preferably 1 up to and including 6, in particular 1 up to and including 4, especially 1 or 2, carbon atoms.

Alkyl - as a group per se and as a structural element of other groups and compounds, such as of halogenoalkyl, alkoxy, alkylthio, alkylsulfinyl, alkylsulfonyl, alkylcarbonyl, alkoxycarbonyl, halogenoalkoxycarbonyl, alkylaminocarbonyl, alkoxyiminomethyl, alkylaminothiocarbonyl and alkylamino - is, in each case taking into due consideration the number, included from case to case, of carbon atoms contained in the corresponding group or compound, either straight-chain, i.e. methyl, ethyl, propyl, butyl, pentyl or hexyl, or branched, for example isopropyl, isobutyl, sec-butyl, tert-butyl, isopentyl, neopentyl or isohexyl.

Alkenyl - as a group per se and as a structural element of other groups and compounds, such as of halogenoalkenyl - is, in each case under due consideration of the number, included from case to case, of carbon atoms contained in the corresponding group or compound, either straight-chain, for example vinyl, 1-methylvinyl, allyl, 1-butenyl or 2-hexenyl, or branched, for example iso-propenyl.

Alkynyl - as a group per se and as a structural element of other groups and compounds, such as of halogenoalkynyl - is, in each case under due consideration of the number, included from case to case, of carbon atoms contained in the corresponding group or compound, either straight-chain, for example propargyl, 2-butynyl or 5-hexynyl, or branched, for example 2-ethynylpropyl or 2-propargylisopropyl.

C₃-C₆cycloalkyl is cyclopropyl, cyclobutyl, cyclopentyl or cyclohexyl.

Alkylene - as a group per se and as a structural element of other groups and compounds, such as of O(alkylene), (alkylene)O, S(=O)_p(alkylene), (alkylene)S(=O)_p or alkylenedioxy - is, in each case under due consideration of the number, included from case to case, of carbon atoms contained in the corresponding group or compound, either straight-chain, for example -CH₂CH₂-, -CH₂CH₂CH₂- or -CH₂CH₂CH₂-, or branched, for example -CH(CH₃)-, -CH(C₂H₅)-, -C(CH₃)₂-, - CH(CH₃)CH₂- or -CH(CH₃)CH(CH₃)-.

Alkenylene is, in each case under due consideration of the number, from case to case, of carbon atoms contained in the corresponding compound, either straight-chain, for example vin-1,2-ylene, all-1,3-ylene, but-1-en-1,4-ylene or hex-2-en-1,6-ylene, or branched, for example 1-methylvin-1,2-ylene.

Alkynylene is, in each case under due consideration of the number, from case to case, of carbon atoms contained in the corresponding compound, either straight-chain, for example propargylene, 2-butynylene or 5-hexynylene, or branched, for example 2-ethynylpropylene or 2-propargylisopropylene.

Aryl is phenyl or naphthyl, in particular phenyl.

Heterocyclyl is a 5- to 7-membered aromatic or non-aromatic ring having one to three heteroatoms, which are selected from the group consisting of N, O and S. 5- and 6-membered rings which contain a nitrogen atom as a heteroatom and, if appropriate, a further heteroatom, preferably nitrogen or sulfur, in particular nitrogen, are preferred.

Halogen - as a group per se and as a structural element of other groups and compounds, such as of halogenoalkyl, halogenoalkenyl and halogenoalkynyl - is fluorine, chlorine, bromine or iodine, especially fluorine, chlorine or bromine, in particular fluorine or chlorine, very especially fluorine.

Halogen-substituted carbon-containing groups and compounds, such as halogenoalkyl, halogenoalkenyl or halogenoalkynyl, can be partly halogenated or perhalogenated, and in the case of polyhalogenation, the halogen substituents can be identical or different. Examples of halogenoalkyl- as a group per se and as a structural element of other groups and compounds, such as of halogenoalkenyl- are methyl which is mono- to trisubstituted by fluorine, chlorine and/or bromine, such as CHF₂ or CF₃; ethyl which is mono- to pentasubstituted by fluorine, chlorine and/or bromine, such as CH₂CF₃, CF₂CF₃, CF₂CCl₃, CF₂CHCl₂, CF₂CHCl₂, CF₂CHCl₅, CF₂CHCl₅; propyl or

isopropyl which is mono- to heptasubstituted by fluorine, chlorine and/or bromine, such as CH₂CHBrCH₂Br, CF₂CHFCF₃, CH₂CF₂CF₃ or CH(CF₃)₂; and butyl or one of its isomers which is mono- to nonasubstituted by fluorine, chlorine and/or bromine, such as CF(CF₃)-CHFCF₃or CH₂(CF₂)₂CF₃. Halogenoalkenyl is, for example, CH₂CH=CHCl, CH₂CH=CCl₂, CH₂CF=CF₂ or CH₂CH=CHCH₂Br. Halogenoalkynyl is, for example, CH₂C≡CF, CH₂C≡CCH₂Cl or CF₂C≡CGH₂F.

Some compounds I to VI and VIII can be present as tautomers, as is familiar to the expert, in particular if AR₇ is H. Compounds I above and below are therefore also to be understood as meaning corresponding tautomers, even if the latter are not mentioned specifically in each case.

Compounds I to VI and VIII which contain at least one basic centre, can form, for example, acid addition salts. These are formed, for example, with strong inorganic acids, such as mineral acids, for example perchloric acid, sulfuric acid, nitric acid, nitrous acid, a phosphoric acid, or a hydrogen halide acid, with strong inorganic carboxylic acids, such as C1-C4alkanecarboxylic acids which are unsubstituted or substituted, for example by halogen, for example acetic acid, such as dicarboxylic acids which are saturated or unsaturated, for example oxalic, malonic, succinic, maleic, fumaric or phthalic acid, such as hydroxycarboxylic acids, for example ascorbic, lactic, malic, tartaric or citric acid, or such as benzoic acid, or with organic sulfonic acids, such as C1-C4alkane- or arylsulfonic acids which are unsubstituted or substituted, for example by halogen, for example methane- or ptoluenesulfonic acid. Compounds I with at least one acid group can furthermore form salts with bases. Suitable salts with bases are, for example, metal salts such as alkali metal or alkaline earth metal salts, for example sodium, potassium or magnesium salts, or salts with ammonia or an organic amine, such as morpholine, pipendine, pyrrolidine, a mono-, di- or tri-lower alkylamine, for example ethyl-, diethyl-, triethyl- or dimethyl-propyl-amine, or a mono-, di- or trihydroxy-lower alkylamine, for example mono-, di- or triethanolamine. Furthermore, where appropriate, corresponding inner salts can be formed. Agrochemically advantageous salts are preferred in the context of the invention; however, salts which have disadvantages for agrochemical uses, for example salts which are toxic to bees or fish, which are employed, for example, for isolation or purification of free compounds I or agrochemically usable salts thereof, are also included, Compounds of the formulae I to VI and VIII in the free form and in the form of their salts are also to be understood above and below as meaning the corresponding salts or the free compounds I to VI and VIII. The same applies to tautomers of compounds of the formulae I to VI and VIII and salts thereof. In general, the free form is in each case preferred.

The reactions described above and below are carried out in a manner known per se, for example in the absence or usually in the presence of a suitable solvent or diluent or a mixture thereof, the reaction being carried out, as required, with cooling, at room temperature or with heating, for example in a temperature range from about -80°C up to the boiling point of the reaction medium, preferably from about 0°C up to about 150°C, and, if necessary, in a closed vessel, under pressure, in an inert gas atmosphere and/or under anhydrous conditions. Particularly advantageous reaction conditions can be seen from the examples.

The starting materials mentioned above and below, which are used for the preparation of the compounds I, in each case in the free form or in salt form, are known or can be prepared by methods known per se, for example in accordance with the following statements.

Variants a1/a2)

Suitable leaving groups X_1 in compounds III are, for example, hydroxyl, C_1 - C_8 alkoxy, halogeno- C_1 - C_8 alkoxy, C_1 - C_8 alkanoyloxy, mercapto, C_1 - C_8 alkylthio, halogeno- C_1 - C_8 alkylthio, C_1 - C_8 alkanesulfonyloxy, halogeno- C_1 - C_8 alkanesulfonyloxy, benzenesulfonyloxy, toluenesulfonyloxy and halogen, preferably toluenesulfonyloxy, trifluoromethanesulfonyloxy and halogen, in particular halogen.

Suitable bases for facilitating the reaction are, for example, alkali metal or alkaline earth metal hydroxides, hydrides, amides, alkanolates, acetates, carbonates, dialkylamides or alkylsilylamides, alkylamines, alkylenediamines, N-alkylated or non-alkylated, saturated or unsaturated cycloalkylamines, basic heterocyclic compounds, ammonium hydroxides and carbocyclic amines. Examples are sodium hydroxide, hydride, amide, methanolate, acetate and carbonate, potassium tert-butanolate, hydroxide, carbonate, and hydride, lithium diisopropylamide, potassium bis(trimethylsilyl)amide, calcium hydride, triethylamine, diisopropyl-ethyl-amine, triethylenediamine, cyclohexylamine,

N-cyclohexyl-N,N-dimethyl-amine, N,N-diethylaniline, pyridine,

4-(N,N-dimethylamino)pyridine, quinuclidine, N-methylmorpholine,

benzyl-trimethyl-ammonium hydroxide and 1,5-diazabicyclo[5.4.0]undec-5-ene (DBU).

The reaction partners can be reacted with one another as such, i.e. without addition of a solvent or diluent, for example in the melt. However, the addition of an inert solvent or diluent or of a mixture thereof is usually advantageous. Examples of such solvents or diluents are: aromatic, aliphatic and alicyclic hydrocarbons and halogenohydrocarbons. such as benzene, toluene, xylene, mesitylene, tetralin, chlorobenzene, dichlorobenzene, bromobenzene, petroleum ether, hexane, cyclohexane, methylene chloride, chloroform, carbon tetrachloride, dichloroethane, trichloroethene or tetrachloroethene; esters, such as ethyl acetate; ethers, such as diethyl ether, dipropyl ether, diisopropyl ether, dibutyl ether, tert-butyl methyl ether, ethylene glycol monomethyl ether, ethylene glycol monoethyl ether, ethylene glycol dimethyl ether, dimethoxydiethyl ether, tetrahydrofuran or dioxane; ketones, such as acetone, methyl ethyl ketone or methyl isobutyl ketone; alcohols, such as methanol, ethanol, propanol, isopropanol, butanol, ethylene glycol or glycerol; amides, such as N,N-dimethylformamide, N,N-diethylformamide, N,N-dimethylacetamide, N-methylpyrrolidone or hexamethylphosphoric acid triamide; nitriles such as acetonitrile or propionitrile; and sulfoxides, such as dimethyl sulfoxide. If the reaction is carried out in the presence of a base, bases employed in excess, such as triethylamine, pyridine, N-methylmorpholine or N,N-diethylaniline, can also serve as the solvent or diluent. The reaction is advantageously carried out in a temperature range from about 0°C up to

The reaction is advantageously carried out in a temperature range from about 0°C up to about 180°C, preferably from about 10°C up to about 80°C, in many cases in the range between room temperature and the reflux temperature of the reaction mixture.

The reaction is preferably carried out under normal pressure.

The reaction can be carried out without an inert gas atmosphere; preferably, however, it is carried out under an inert gas atmosphere, for example nitrogen or argon, in particular nitrogen.

The reaction time is not critical; a reaction time of about 0.1 to about 24 hours, in particular about 0.5 to about 2 hours, is preferred.

The product is isolated by customary methods, for example by filtration, crystallization, distillation or chromatography or any suitable combination of these processes.

In a preferred embodiment of variants a1/a2), a compound II is reacted with a compound III at 0°C to 80°C, preferably 10°C to 30°C, in an inert solvent, preferably an amide, in

particular N,N-dimethylformamide in the presence of a metal hydride, preferably sodium hydride.

Particularly preferred conditions for the reaction are described in Examples H1d) and H 3f).

The compounds of the formula III are known or can be prepared analogously to known compounds.

The compounds I are known. However, their preparation according to the prior art has a large number of serious industrial, ecological, economic and other disadvantages.

Thus, in the preparation processes according to the prior art, as a rule E/Z isomer mixtures with respect to the C=N double bond marked with E in formula I are obtained. Since the biological properties of the E isomers are in each case found to be superior to those of the mixtures and of the Z isomers in each case, the processes according to the prior art have the significant disadvantage that products are produced which are either significantly less active as E/Z mixtures or from which the Z isomers must be removed in order to increase their biological activity, which means that many unnecessary handling operations must be carried out for separation of isomers, which has the effect of being very time-consuming. blocks valuable production lines for a long time and is associated with high additional energy costs. The removal of the less active Z isomer also leads to additional enormous losses in yield, which in turn not only is problematic and ecologically disadvantageous, but also renders the process according to the prior art much more expensive and consequently economically of no interest. The industrial, ecological, economic and other disadvantages of the processes according to the prior art are not limited to those described above, these latter being intended to serve only as a few examples of the large number of disadvantages of the processes according to the prior art. The disadvantages of the processes according to the prior art cause serious problems even when the processes are carried out on a laboratory scale. When the processes are carried out on a larger scale, these disadvantages intensify considerably. In the end, however, the aim is to carry out a specific process on an industrial scale if this process is to be suitable for preparing products for agrochemical purposes.

According to the process of the present invention, the compounds I are prepared by reaction of the compound II with a compound III or by reaction of the compound IV with a compound V. These processes according to the invention have extremely surprising

industrial, ecological, economic and other advantages compared with the processes from the prior art. Since the compounds II or, respectively, IV are present in the preparation process according to the invention as pure E isomers in respect of the C=N double bond marked with E, only the E isomer of the compounds I is produced in the present process, which has the effect of an enormous saving in time and at the same time a high saving in cost and energy, since no valuable production lines are blocked for a long time for separation of the isomers, and at the same time the amount of biologically more active E isomer produced by per unit time is much higher than in the processes according to the prior art. The resources such as starting products and energy are consequently utilized to the optimum in the present process, which not only very greatly simplifies the process and renders it ecologically advantageous, but consequently renders it cheaper and therefore of greater economic interest. This means that all the disadvantages of the processes according to the prior art which can be attributed to the formation of E/Z isomers are avoided. The industrial, ecological, economic and other advantages of the process according to the invention are not limited only to those described above, these latter being intended to serve only as a few examples of the large number of advantages inherent in this process. Due to all the abovementioned advantages of the present process, the serious problems which occur in the processes according to the prior art are avoided even at the stage of a laboratory process. If the present process is used on a larger scale, these advantages prove to be even much more significant, which has the effect that these advantages first allow the process to be used on an industrial scale.

For this reason, all the industrial, ecological, economic and other disadvantages of the processes according to the prior art are surprisingly advantageously overcome in the preparation of compounds I by the present process.

Variant b)

The process according to variant b) is carried out by first reacting compound VI with compound VII, if appropriate further reacting the resulting product IV, if appropriate after isolation, with hydroxylamine or a salt thereof, and further reacting the resulting products II or, respectively, IV, if appropriate after isolation, in accordance with variants a1/a2), for example in the manner described above, to give the compounds I.

Suitable leaving groups X_2 in the compounds VII are, for example, those which are mentioned as examples for X_1 in variants a1/a2).

Suitable bases for facilitating the reaction are, for example, those which are mentioned in variants a1/a2).

The reaction partners can be reacted with one another as such, i.e. without addition of a solvent or diluent, for example in the melt. However, the addition of an inert solvent or diluent or of a mixture thereof is usually advantageous. Examples of such solvents or diluents are those mentioned in variants a1/a2).

The reaction is advantageously carried out in a temperature range from about 0°C to about 180°C, preferably from about 10°C to about 80°C, in many cases in the range between room temperature and the reflux temperature of the reaction mixture.

The reaction is preferably carried out under normal pressure.

The reaction can be carried out without an inert gas atmosphere; preferably, however, it is carried out under an inert gas atmosphere, for example nitrogen or argon, in particular nitrogen.

The reaction time is not critical; a reaction time of about 0.1 to about 24 hours, in particular about 0.5 to about 5 hours, is preferred.

The product is isolated by customary methods, for example filtration, crystallization, distillation or chromatography or any suitable combination of these processes.

In a preferred embodiment of variant b), a compound VI is reacted with a compound VII at 0°C to 80°C, preferably 10°C to 60°C, in an inert solvent, preferably a nitrile, in particular acetonitrile, in the presence of a metal carbonate, preferably potassium carbonate, and the compound IV thus obtainable is then further reacted, preferably in accordance with method a2).

Particularly preferred conditions for the reaction are described in Examples H 1b) to 1d) and H 3d) to 3f).

The compounds of the formula VII are known or can be prepared analogously to known compounds.

The present process according to the invention of variant b), which in principle is an advantageous combination of an O-alkylation reaction with process variants a1/a2)

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according to the invention, has all the great advantages compared with the prior art which have already been discussed above for the process according to the invention of variants a1/a2). In particular, the process of variant b) ensures that the E configuration of the C=N double bond marked with E in compound VI is retained. Furthermore, however, the process according to the invention of variant b) also has further industrial, ecological, economic and other advantages which are connected with the specific property that the intermediate product IV initially formed is not purified but is directly further processed as the moist crude product, in the case of intermediate isolation, or in situ in the reaction mixture, if it is not isolated. This missing purification step on the intermediate product mentioned is of advantage, for example, in as much as it is not necessary to dry it, which not only saves energy and further resources, but also enormously increases the safety of the preparation process, since the possible danger of a dust explosion of the dry intermediate product is averted completely. The savings in resources are even greater if the intermediate product is further reacted without purification, since, for example, no additional solvents are consumed for the recrystallization. The process of variant b) is of particular advantage compared with the individual process steps of the alkylation reaction of variants a1/a2) carried out in that the total reaction time in the process of variant b) is much shorter, which consequently leads to a much higher production of reaction product I per unit time and therefore to a much more efficient utilization of the valuable production lines. Furthermore, the total yield of reaction product I is surprisingly good when the process of variant b) is employed, and, compared with the combined yields of the individual process steps of the alkylation reaction and variants a1/a2) carried out, is in the same percentage range or even beffer. The industrial, ecological, economic and other advantages of the process according to the invention of variant b) are not limited to those described above, these latter being intended to serve only as a few examples of the large number of advantages inherent in the process according to the invention of variant b).

By using process variant b) according to the invention for preparation of the compounds I, a large number of industrial, ecological, economic and other advantages can therefore surprisingly be utilized efficiently.

Variant c)

The process according to variant c) is carried out by first reacting compound VIII with an alkylnitrite and further reacting the resulting product VI, if appropriate after isolation, in

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accordance with variant b), for example in the manner described above, to give the compounds I.

Suitable bases for facilitating the reaction are, for example, those which are mentioned in variants a1/a2).

The reaction partners can be reacted with one another as such, i.e. without addition of a solvent or diluent, for example in the melt. However, the addition of an inert solvent or diluent or of a mixture thereof is usually advantageous. Examples of such solvents or diluents are those mentioned in variants a1/a2).

The reaction is advantageously carried out in a temperature range from about 0°C to about 180°C, preferably from about 0°C to about 60°C, in many cases in the range between room temperature and the reflux temperature of the reaction mixture.

The reaction is preferably carried out under normal pressure.

The reaction can be carried out without an inert gas atmosphere; preferably, however, it is carried out under an inert gas atmosphere, for example nitrogen or argon, in particular nitrogen.

The reaction time is not critical; a reaction time of about 0.1 to about 24 hours, in particular about 0.5 to about 3 hours, is preferred.

The product is isolated by customary methods, for example filtration, crystallization, distillation or chromatography or any suitable combination of these processes.

In a preferred embodiment of variant c), a compound VIII is reacted with an alkyl nitrite at 0°C to 80°C, preferably 0°C to 40°C, in an inert solvent, preferably an alcohol, in particular methanol, in the presence of a metal alcoholate, preferably sodium methanolate, and the compound VI thus obtainable is then further reacted, preferably in accordance with method b).

Particularly preferred conditions for the reactions are described in Examples H 3d) to 3f).

The compounds of the formula VIII are known or can be prepared analogously to known compounds.

The present process according to the invention of variant c), which in principle is an advantageous combination of an oximation reaction with process variants a1/a2) and b)

according to the invention, has all the great advantages compared with the prior art which have already been discussed above for the processes according to the invention of variants a1/a2) and b). Furthermore, the present oximation process for the preparation of the compounds VI surprisingly result exclusively in the E configuration of the C=N double bond marked with E in formula VI. It is thus ensured that the particular starting products II, IV or, respectively, VI in the subsequent processes according to the invention for the preparation of the compounds I, for example in process variants a1/a2) and b), are pure E isomers.

A large number of industrial, ecological, economic and other advantages can therefore surprisingly be utilized efficiently by using process variants c) according to the invention for the preparation of the compounds of the formula I.

The E isomers of the compounds of the formulae II, IV and VI and tautomers thereof, in each case in the free form or in salt form, are novel and the present invention likewise relates to them.

The present invention furthermore relates to a process for the preparation of the E isomers of a compound of the formula VI or of a tautomer thereof, in each case in the free form or in salt form, according to the abovementioned process c),

a process for the preparation of the E isomers of a compound of the formula IV, or of a tautomer thereof, in each case in the free form or in salt form, according to the abovementioned process b1), and

a process for the preparation of the E isomers of a compound of the formula II, or of a tautomer thereof, in each case in the free form or in salt form, according to the abovementioned process b2).

The process conditions for the preparation of these intermediate products can be seen from the abovementioned processes a), b) and c).

Preparation Examples

Example H1: Methyl 2-[[(1-methyl-2-phenyl-2-E-[(2-propynyl)oxyimino]ethylidene)amino]oxy]methyl]α-(methoxymethylene)-phenylacetate (Compound 1.16)

H1a) 1-Phenyl-1,2-propanedione 1-E-oxime

69.7 g of a 30 % solution of sodium methylate in methanol are added dropwise to a solution of 40.2 g of 1-phenyl-2-propanone and 36.1 g of isopentyl nitrite in 460 ml of methanol at 20-25°, while cooling. The reaction mixture is then further stirred at room temperature for 1 hour. After the solution has been concentrated in vacuo, the residue is dissolved in 600 ml of water, the solution is acidified with 10 % hydrochloric acid, the product which precipitates out is filtered off and dissolved in ethyl acetate and the organic phase is washed twice with water, dried with sodium sulfate and evaporated in vacuo. The residue is stirred up in hexane and filtered. The title product is thus obtained with a melting of 168-70°C.

H1b) 1-Phenyl-1,2-propanedione 1-E-[(2-propynyl)oxime]

A mixture of 14 g of 1-phenyl-1,2-propanedione 1-E-oxime, 11.9 g of 1-bromo-2-propyne, 13.8 g of potassium carbonate and 0.5 g of potassium iodide in 170 ml of acetonitrile is stirred at 50° for 2 hours, the solvent is then distilled off in vacuo and the residue is dissolved again in ethyl acetate. The organic phase is washed in each case twice with water and saturated sodium chloride solution, dried with sodium sulfate and evaporated in vacuo. After recrystallization of the residue from hexane, 1-phenyl-1,2-propanedione 1-E-[(2-propynyl)oxime] is obtained with a melting point of 54-56°C.

H1d) 1-Phenyl-1,2-propanedione 1-E-[(2-propynyl)oxime]-2-oxime

A mixture of 14.3 g of 1-phenyl-1,2-propanedione 1-E-[(2-propynyl)oxime], 10.3 g of hydroxylamine hydrochloride and 11.7 g of pyridine in 230 ml of ethanol is boiled under reflux for 1 hour and then concentrated in vacuo, and 800 ml of water are added to the residue. The product which has precipitated out is filtered off and dissolved in ethyl acetate and the solution is washed three times with water, dried with sodium sulfate and evaporated in vacuo. The residue is suspended in hexane and filtered. The title product is thus obtained with a melting point of 163-165°C.

H1e) Methyl 2-[[[(1-methyl-2-phenyl-2-E-[(2-propynyl)oxyimino]ethy-lidene)amino]oxy]-methyl]- α -(methoxymethylene)-phenylacetate

A solution of 5 g of 1-phenyl-1,2-propanedione 1-E-[(2-propynyl)oxime]-2-oxime in 24 ml of N,N-dimethylformamide is added dropwise to a suspension of 1.16 g of sodium hydride (about 55% in oil) in 45 ml of N,N-dimethylformamide at room temperature and the mixture is further stirred for 10 minutes. 6.5 g of methyl 2-(bromomethyl)- α -(methoxymethylene)-phenylacetate in 24 ml of N,N-dimethylformamide are then added dropwise and the reaction mixture is further stirred at room temperature for 1 hour. Thereafter, the mixure is acidified

with acetic acid and evaporated in vacuo. The residue is dissolved in ethyl acetate and the solution is washed three times with water and twice with saturated sodium chloride solution, dried with sodium sulfate and evaporated in vacuo. After recrystallization of the residue from hexane/ethyl acetate, the title compound is obtained with a melting point of 82-84°.

Example H2: Methyl 2-[[[(1-methyl-2-(4-fluorophenyl)-2-E-[(2-propynyl)oxyimino]ethylidene)amino]oxy]methyl]- α -(methoxymethylene)-phenylacetate (compound 1.44)

The title compound with a melting point of 91-93° can be prepared in a manner analogous to that described in Example H1, starting from 1-(4-fluorophenyl)-2-propanone.

<u>Example H3</u>: Methyl 2-[[[(1-methyl-2-(4-(3-trifluoromethylphenylmethoxy)-phenyl) 2-E-[(2-propynyl)oxyimino]ethylidene)amino]oxy]methyl]- α -(methoxymethylene)-phenylacetate (compound 1.240)

H3a) 1-(4-Hydroxyphenyl)-2-propanone

A mixture of 82 g of 1-(4-methoxyphenyl)-2-propanone, 500 ml of acetic acid and 500 ml of aqueous hydrobromic acid is boiled under reflux for 2 hours and then evaporated in vacuo. The oily residue is extracted four times with 700 ml of hexane/ether (5:2) each time, the extract is evaporated and the residue is chromatographed over silica gel using hexane/ethyl acetate (3:1). 1-(4-Hydroxyphenyl)-2-propanone is thus obtained with a melting point of 40-41°.

H3b) 1-f4-(3-Trifluoromethylphenylmethoxy)-phenyl]-2-propanone

A mixture of 5.8 g of 1-(4-hydroxyphenyl)-2-propanone, 61.6 g of potassium carbonate, 72.3 g of 1-(chloromethyl)-3-(trifluoromethyl)-benzene, and 1 g of potassium iodide in 800 ml of acetone is boiled under reflux for 5 hours. Thereafter, the reaction mixture is filtered and the filtrate is evaporated in vacuo. The residue is then dissolved in diethyl ether and the ethereal phase is washed three times with water, dried with sodium sulfate and evaporated. The 1-[4-(3-trifluoromethylphenylmethoxy)-phenyl]-2-propanone thus obtainable is employed in the next reaction stage without further purification.

H3c) 1-[4-(3-Trifluoromethylphenylmethoxy)-phenyl]-1,2-propanedione 1-E-oxime 45 g of a 30% solution of sodium methanolate in methanol are slowly added dropwise to a solution of 59.6 g of 1-[4-(3-trifluoromethylphenylmethoxy)-phenyl]-2-propanone and 23.4 g of isopentyl nitrite in 300 ml of methanol such that the temperature does not exceed 20-25°. The reaction mixture is then further stirred at room temperature for 1 hour and thereafter evaporated in vacuo. The residue is dissolved in 600 ml of water and the solution is

acidified with 10% hydrochloric acid. The precipitate which separates out is filtered off and dissolved in ethyl acetate and the organic phase is washed twice with water, dried with sodium sulfate and evaporated. After the crude product has been suspended in hexane and filtered, 1-[4-(3-trifluoromethylphenylmethoxy)-phenyl]-1,2-propanedione 1-E-oxime is obtained with a melting point of 134-136°.

H3d) 1-[4-(3-Trifluoromethylphenylmethoxy)-phenyl]-1,2-propanedione 1-E-[(2-propynyl)oxime]

A mixture of 6 g of 1-[4-(3-trifluoromethylphenylmethoxy)-phenyl]-1,2-propanedione 1-E-oxime, 2.4 g of 1-bromo-2-propyne, 2.6 g of potassium carbonate and 0.5 g of potassium iodide in 40 ml of acetonitrile is boiled under reflux for 1 hour and then evaporated in vacuo and the residue is dissolved in ethyl acetate. The organic phase is washed twice with water and once with saturated sodium chloride solution, dried with sodium sulfate and evaporated. The crude 1-[4-(3-trifluoromethylphenylmethoxy)-phenyl]-1,2-propanedione 1-E-[(2-propynyl)oxime] thus obtainable is further processed without further purification.

H3e) 1-[4-(3-Trifluoromethylphenylmethoxy)-phenyl]-1,2-propanedione 1-E-[(2-propynyl)oxime]-2-oxime

A mixture of 5.9 g of 1-[4-(3-trifluoromethylphenylmethoxy)-phenyl]-1,2-propanedione 1-E-[(2-propynyl)oxime], 2.3 g of hydroxylamine hydrochloride and 2.6 g of pyridine in 60 ml of ethanol is boiled under reflux for 1 hour and then concentrated in vacuo, and 200 ml of water are added to the residue. The product which has precipitated out is filtered off and dissolved in ethyl acetate, and the solution is washed twice with water and once with saturated sodium chloride solution, dried with sodium sulfate and evaporated in vacuo. The residue is suspended in hexane and filtered. 1-[4-(3-Trifluoromethylphenylmethoxy)-phenyl]-1,2-propanedione 1-E-[(2-propynyl)oxime]-2-oxime is thus obtained with a melting point of 114-115°.

H3f) Methyl 2-[[[(1-methyl-2-(4-(3-trifluoromethylphenylmethoxy)-phenyl)-2-É-[(2-propynyl)oxyimino]ethylidene)amino]oxy]methyl]- α -(methoxymethylene)-phenylacetate A solution of 5.5 g of 1-[4-(3-trifluoromethylphenylmethoxy)-phenyl]-1,2-propanedione 1-E-[(2-propynyl)oxime]-2-oxime in 25 ml of N,N-dimethylformamide is added dropwise to a suspension of 0.7 g of sodium hydride (about 55% in oil) in 25 ml of N,N-dimethylformamide and the mixture is further stirred at room temperature for 10 minutes. 4 g of methyl 2-(bromomethyl)- α -(methoxymethylene)-phenylacetate in 15 ml of N,N-dimethylformamide are

then added dropwise and the reaction mixture is further stirred at room temperature for 1 hour. Thereafter, the mixture is acidified with acetic acid and evaporated in vacuo at 50°. The residue is dissolved in ethyl acetate and the solution is washed twice with water and once with saturated sodium chloride solution, dried with sodium sulfate and evaporated in vacuo. After purification by chromatography (silica gel, ethyl acetate/hexane 1:3), the title compound is obtained as a resin.

Example H4: Methyl 2-[[[(1-methyl-2-(4-(4-chlorophenoxy)-phenyl)-2-E-[(2-ethyl)oxyimino]ethylidene)amino]oxy]methyl]- α -(methoxymethylene)-phenylacetate (compound 1.366)

H4a) 1-[4-(4-Chlorophenoxy)-phenyl]-1,2-propanedione 1-E-oxime

16.7 g of a 30% solution of sodium methylate in methanol are added dropwise to a solution of 22.5 g of 1-[4-(4-chlorophenoxy)-phenyl]-2-propanone and 10.3 g of isopentyl nitrite in 120 ml of methanol at 20-25°, while cooling. The reaction mixture is then further stirred at room temperature for 1 hour. After the solution has been concentrated in vacuo, the residue is dissolved in 300 ml of water and the solution is acidified with 10% hydrochloric acid, the product which precipitates out is filtered off and dissolved in ethyl acetate and the organic phase is washed twice with water, dried with sodium sulfate and evaporated in vacuo. The residue is stirred up in hexane and filtered. The title product is thus obtained with a melting point of 154-155°C.

H4b) 1-[4-(4-Chlorophenoxy)-phenyl]-1,2-propanedione 1-E-[(2-ethyl)oxime]

A mixture of 6 g of) 1-[4-(4-Chlorophenoxy)-phenyl]-1,2-propanedione 1-E-oxime, 3.3 g of ethyl bromide, 3.5 g of potassium carbonate and 0.5 g of potassium iodide in 30 ml of acetonitrile is stirred at 50° for 2 hours, the solvent is then distilled off in vacuo and the residue is dissolved again in ethyl acetate. The organic phase is washed in each case twice with water and saturated sodium chloride solution, dried with sodium sulfate and evaporated in vacuo. After recrystallization of the residue from hexane, the title product is obtained with a melting point of 77-78°C.

H4c) 1-[4-(4-Chlorophenoxy)-phenyl]-1,2-propanedione 1-E-[(2-ethyl)oxime]-2-oxime

A mixture of 5.5 g of 1-[4-(4-chlorophenoxy)-phenyl]-1,2-propanedione 1-E-[(2-ethyl)oxime],

2.4 g of hydroxylamine hydrochloride and 2.7 g of pyridine in 50 ml of ethanol is boiled
under reflux for 1 hour and then concentrated in vacuo, and 800 ml of water are added to
the residue. The product which has precipitated out is filtered off and dissolved in ethyl

acetate and the solution is washed three times with water, dried with sodium sulfate and evaporated in vacuo. The residue is suspended in hexane and filtered. The title product is thus obtained in a pure form with a melting point of 176-177°C.

H4d) Methyl 2-[[[(1-methyl-2-(4-(4-chlorophenoxy)-phenyl)-2-E-[(2-ethyl)oxyimino]ethylidene)amino]oxy]methyl]- α -(methoxymethylene)-phenylacetate.

A solution of 4.7 g of 1-[4-(4-chlorophenoxy)-phenyl]-1,2-propanedione 1-E-[(2-propynyl)-oxime]-2-oxime in 25 ml of N,N-dimethylformamide is added dropwise to a suspension of 0.65 g of sodium hydride ($\frac{1}{4}$ bout 55% in oil) in 20 ml of N,N-dimethylformamide and the mixture is further stirred at room temperature for 10 minutes. 4 g of methyl 2-(bromomethyl)- α -(methoxymethylene)-phenylacetate in 15 ml of N,N-dimethylformamide are then added dropwise and the reaction mixture is further stirred at room temperature for 1 hour. Thereafter, the mixture is acidified with acetic acid and evaporated in vacuo at 50°. The residue is dissolved in ethyl acetate and the solution is washed twice with water and once with saturated sodium chloride solution, dried with sodium sulfate and evaporated in vacuo. After purification by flash chromatography (silica gel, ethyl acetate/hexane 1:3), the title compound is obtained with a melting point of 87-89°C.

Example H5: Methyl 2-[[(1-methyl-2-(4-(4-chlorophenoxy)-phenyl)-2-E-[(2-ethyl)oxy-imino]ethylidene)amino]oxy]methyl]- α -(methoxyimino)-phenylacetate (compound 2.366)

The title compound with a melting point of 90 to 93°C is obtained in a manner analogous to that described in Example H4 from 1-[4-(4-chlorophenoxy)-phenyl]-1,2-propanedione 1-E-[(2-propynyl)oxime]-2-oxime and methyl 2-(bromomethyl)- α -(methoxyimino)-phenylacetate.

Example H6: 2-[[(1-Methyl-2-(4-(4-chlorophenoxy)-phenyl)-2-E-[(2-ethyl)oxy-imino]ethylidene)amino]oxy]methyl]- α -(methoxyimino)-phenylacetic acid methylamide (compound 3.366)

13,3 g of methyl 2-[[[(1-methyl-2-(4-(4-chlorophenoxy)-phenyl)-2-E-[(2-ethyl)oxyimino]ethyli-ene)amino]oxy]methyl]-α-(methoxyimino)-phenylacetate are left to stand together with 80 ml of dimethylformamide and 9.2 ml of an 8 molar solution of methylamine in ethanol at room temperature for two days. The mixture is concentrated at 50°C, n-hexane is added and the mixture is cooled to room temperature and filtered. The residue is dried under a high vacuum. The title compound is obtained with a melting point of 126-129°C.

Example H7: The other compounds listed in Tables 1 to 3 can also be prepared in a manner analogous to that described in Examples H1 to H6. In the "physical data" column of the tables, the temperatures stated in each case designate the melting point of the compound in question. c.propyl is cyclopropyl.

Table 1

Compounds of the general formula

$$CH_3O$$
 CH_3
 CH_3

in which X is CH and Y is oxygen and the combination of substituents R_2 , $(R_5)_n$ and A-R₇ for a compound in each case corresponds to a line in Table A. The compound numbers of the following table correspond to the particular numbers in Table A.

Compound No.	² .	Phys. Data(Melting point °C)
1.14		75-77°
1.16		82-84°
1.22	1	111-113° ;
1.42	٠	Resin
1.44		91-93°
1.50		Resin
1.70		Resin
1.72		Resin
1.78		Resin
1.225		102-103°
1.226		81-83°
1.227		Resin
1.233	;	Resin
1.234		73-75°
1.238		Resin
1.240	•	Resin
1.241		Resin
1.242		Resin
1.244	•	Resin
1.245	4	Resin
1.294		Resin
1.296		112-114°
1.366		87-89°

<u>Table 2</u>
Compounds of the general formula I.1, in which

X is nitrogen and

Y is oxygen

and the combination of substituents R_2 , $(R_5)_n$ and A- R_7 for a compound in each case corresponds to a line in Table A.

Compound No.	;	Melting point (°C)	
2.198	·····	75-77	
2.254		80-82	
2.309		106-108	
2.310		102-104	
2.366		90-93	

Table 3

Compounds of the general formula I.1, in which

X is nitrogen and

Y is NH and

the combination of substituents R_2 , $(R_5)_n$ and $A-R_7$ for a compound in each case corresponds to a line in Table A.

Compound No.		Melting point (°C)	
	3.198	75-77	
	3.254	112-114	
	3.309	89-91	
	3.310	88-90	
	3.366	126-129	

Table A

Compound No.	R₂	(R₅)n	A-R ₇
1	CH₃	Н	CH₃
2	CH ₃	н	°C₂H₅
3	CH3	Н	n-C₃H ₇

Compound	R ₂	(R ₅) _n	A-R ₇
No.		*.	
4	СН₃	Н	i-C₃H ₇
5	CH₃	Н	n-C₄H ₉
6	CH₃	H	n-C ₆ H ₁₃
7	CH₃	Н	CH₂F
8	CH3	Н	CHF₂
9	CH ₃	Н	CH ₂ CF ₃
10	CH ₃	Н	CH ₂ CH=CH ₂
11	CH ₃	Н	CH₂CH=CHCH₃
12	CH₃	Н	CH ₂ CH=C(CH ₃) ₂
13	CH₃	Н	CH₂CH=CHCI
1.4	CH₃	Н	CH2CH=CCI2
15	CH₃	Н	CH ₂ C(CH ₃)=CH ₂
16	CH₃	H	CH₂C≡CH
17	CH ₃	· H .	CH₂Si(CH₃)₃
18	CH ₃	Н	CH ₂ -c.propyl-2,2-Cl ₂
19	CH₃	Н	CH₂CN
20	CH₃	Н	CH₂COOC₂H₅
21	CH₃	Н	CH(CH₃)COOC₂H₅
22	CH ₃	Н	CH ₂ C ₆ H ₄ -3-CF ₃
23	CH₃	Н	CH₂C ₆ H₄-4-F
24	CH₃	H	CH ₂ C ₆ H ₄ -3-F
25	CH ₃	Н	CH ₂ C ₆ H ₄ -2-F
26	CH ₃	н	C(=O)OC₂H₅
27	CH ₃	н	C(=O)NHCH₃
28	CH ₃	н	C(=0)C(=0)OC ₂ H ₅
29	CH₃	4-F	CH₃
30	CH₃	4-F	C ₂ H ₅
31	CH3	4-F	n-C₃H ₇
32	CH₃	4-F	i-C ₃ H ₇
33	CH ₃	4-F	n-C₄H ₉

Compound	R ₂	(R₅) _n	A-R ₇
No.	,	,	
34	CH₃	4-F	n-C ₆ H ₁₃
35	CH₃	4-F	CH₂F
36	CH₃	4-F	CHF ₂
37	CH₃	4-F	CH ₂ CF ₃
38	CH ₃	4-F	CH ₂ CH=CH ₂
39	CH₃	4-F	CH2CH=CHCH3
40	CH₃	4- F	CH ₂ CH=C(CH ₃) ₂
41	CH₃	4-F	CH₂CH=CHCI
42	CH₃	4-F	CH₂CH=CCI₂
43	CH₃	4-F	CH ₂ C(CH ₃)=CH ₂
44	CH₃	4-F	CH₂C≡CH
45	CH₃	4-F	CH ₂ Si(CH ₃) ₃
46	СН₃	4-F	CH ₂ -c.propyl-2,2-Cl ₂
47	СН₃	4-F	CH₂CN
48	CH₃	4-F	CH₂COOC₂H₅
49	СН₃	4-F	CH(CH₃)COOC₂H₅
50	CH₃	4-F	CH ₂ C ₆ H ₄ -3-CF ₃
51	CH₃	4-F	CH₂C ₆ H₄-4-F
52	CH₃	4-F	CH₂C ₆ H₄-3-F
53	CH₃	4-F	CH₂C ₆ H₄-2-F
54	CH ₃	4-F	C(=O)OC ₂ H ₅
55	CH₃	4-F	C(=O)NHCH ₃
56	CH₃	4-F	C(=O)C(=O)OC ₂ H ₅
57	CH₃	4-OCH₃	CḤ₃
58	CH₃	4-OCH ₃	C ₂ H ₅
59	CH₃	4-OCH₃	n-C ₃ H ₇
60	CH₃	4-OCH₃	i-C ₃ H ₇
61	CH₃	4-OCH ₃	n-C₄H ₉
62	CH₃	4-OCH₃	n-C ₆ H ₁₃
63	CH₃	4-OCH ₃	CH₂F

Compound	R ₂	(R ₅) _n	A-R ₇
No.			
64	CH₃	4-OCH₃	CHF₂ .
65	CH₃	4-OCH₃	CH₂CF₃
66	CH ₃	4-OCH3	CH ₂ CH=CH ₂
67	CH₃	4-OCH ₃ .	CH₂CH=CHCH₃
68	CH ₃	4-OCH ₃	CH ₂ CH=C(CH ₃) ₂
69	CH₃	4-OCH ₃	CH₂CH=CHCI
70	CH ₃	4-OCH ₃	CH2CH=CCl2
71	CH₃	4-OCH ₃	CH ₂ C(CH ₃)=CH ₂
72	CH ₃	4-OCH₃	CH₂C≘CH
73	CH₃	4-OCH₃	CH₂Si(CH₃)₃
74	СН₃	4-OCH ₃	CH ₂ -c.propyl-2,2-Cl ₂
75	CH₃	4-OCH₃	CH₂CN
76	CH ₃	4-OCH ₃	CH₂COOC₂H₅
77	CH₃	4-OCH ₃	CH(CH ₃)COOC ₂ H ₅
78	CH₃	4-OCH ₃	CH ₂ C ₆ H ₄ -3-CF ₃
79	CH₃	4-OCH₃	CH₂C ₆ H₄-4-F
80	CH ₃	4-OCH₃	CH₂C ₆ H₄-3-F
81	CH₃	4-OCH₃	CH₂C ₆ H₄-2-F
82	CH ₃	4-OCH₃	C(=0)OC ₂ H ₅
83	CH₃	4-OCH₃	C(=O)NHCH₃
84	CH₃	4-OCH ₃	$C(=O)C(=O)OC_2H_5$
85	CH₃	4-OC ₂ H ₅	CH ₃
86	CH₃-	4-OC ₂ H ₅	C₂H₅
87	CH₃	4-OC ₂ H ₅	n-C ₃ H ₇
88	CH ₃	4-OC₂H₅	i-C ₃ H ₇
89	CH ₃	4-OC₂H₅	n-C ₄ H ₉
90	CH ₃	4-OC₂H₅	n-C ₆ H ₁₃
91	CH₃	4-OC₂H₅	CH₂F
92	CH₃	4-OC ₂ H ₅	CHF₂
93	CH ₃	4-OC₂H₅	[∞] €H₂CF₃

Compound	R ₂	(R ₅) _n	A-R ₇
No.	2		
94	CH₃	4-OC₂H₅	CH ₂ CH=CH ₂
95	CH₃	4-OC₂H₅	CH₂CH=CHCH₃
96	CH₃	4-OC ₂ H ₅	CH ₂ CH=C(CH ₃) ₂
97	CH₃	4-OC₂H₅	CH₂CH=CHCI
98	CH ₃	4-OC ₂ H ₅	CH₂CH=CCl₂
99	CH ₃	4-OC ₂ H ₅	CH ₂ C(CH ₃)=CH ₂
100	CH₃	4-OC₂H₅	CH₂C≡CH
101	CH₃	4-OC ₂ H ₅	CH ₂ Si(CH₃)₃
102	CH ₃	4-OC₂H₅	CH₂-c.propyl-2,2-Cl₂
103	CH ₃	4-OC ₂ H ₅	CH₂CN
104	CH ₃	4-OC ₂ H ₅	CH ₂ COOC ₂ H ₅
105	CH ₃	4-OC₂H₅	CH(CH₃)COOC₂H₅
106	CH ₃	4-OC ₂ H ₅	CH ₂ C ₆ H ₄ -3-CF ₃
107	CH₃	4-OC ₂ H ₅	CH₂C ₆ H₄-4-F
108	CH₃	4-OC ₂ H ₅	CH₂C ₆ H₄-3-F
109	CH₃	4-OC ₂ H ₅	CH ₂ C ₆ H ₄ -2-F
110	CH₃	4-OC₂H₅	C(=O)OC ₂ H ₅
111	CH₃	4-OC₂H₅	C(=O)NHCH₃
112	CH ₃	4-OC₂H₅	C(=O)C(=O)OC ₂ H ₅
113	CH₃	4-O-n-C₃H ₇	CH₃
114	CH ₃	4-O-n-C ₃ H ₇	C ₂ H ₅
115	CH₃	4-O-n-C ₃ H ₇	n-C ₃ H ₇
116	CH ₃	4-O-n-C₃H ₇	i-C₃H ₇
117	CH ₃	4-O-n-C ₃ H ₇	n-C₄H ₉
118	CH ₃	4-O-n-C₃H ₇	n-C ₆ H ₁₃
119	CH ₃	4-O-n-C ₃ H ₇	CH₂F
120	CH ₃	4-O-n-C₃H ₇	CHF₂
121	CH₃ :	4-O-n-C₃H ₇	CH₂CF₃
122	CH₃	4-O-n-C₃H ₇	CH₂CH=CH₂
123	CH₃	4-O-n-C₃H ₇	ີ່CH₂CH=CHCH₃

Compound	R ₂	(R ₅) _n	A-R ₇
No.		2 .	
124	CH₃	4-O-n-C₃H ₇	CH₂CH=C(CH₃)₂
125	CH₃	4-O-n-C ₃ H ₇	CH₂CH=CHCI
126	CH ₃	4-O-n-C ₃ H ₇	CH ₂ CH=CCl ₂
127	CH ₃	4-O-n-C ₃ H ₇	CH₂C(CH₃)=CH₂
128	CH ₃	4-O-n-C₃H ₇	CH₂C≘CH
129	CH ₃	4-O-n-C ₃ H ₇	CH ₂ Si(CH ₃) ₃
130	CH₃	4-O-n-C₃H ₇	CH ₂ -c.propyl-2,2-Cl ₂
131	CH₃	4-O-n-C ₃ H ₇	CH₂CN
132	CH₃	4-O-n-C ₃ H ₇	CH₂COOC₂H₅
133	CH ₃	4-0-n-C ₃ H ₇	CH(CH₃)COOC₂H₅
134	CH₃	4-O-n-C ₃ H ₇	CH₂C ₆ H ₄ -3-CF₃
135	CH₃	4-O-n-C ₃ H ₇	CH₂C ₆ H₄-4-F
136	CH₃	4-O-n-C ₃ H ₇	CH ₂ C ₆ H ₄ -3-F
137	CH₃	4-O-n-C ₃ H ₇	CH₂C ₆ H₄-2-F
138	CH₃	4-O-n-C ₃ H ₇	C(=O)OC ₂ H ₅
139	CH₃	4-O-n-C ₃ H ₇	C(=O)NHCH₃
140	CH₃	4-O-n-C ₃ H ₇	C(=0)C(=0)OC ₂ H ₅
141	CH ₃	2-CH₃	CH₃
142	CH₃	2-CH ₃	C₂H₅
143	CH₃	2-CH₃	n-C₃H ₇
144	CH ₃	2-CH₃	i-C₃H ₇
145	CH ₃	2-CH₃	n-C₄H ₉
146	CH ₃	2-CH₃	n-C ₆ H ₁₃
147	CH ₃	2-CH₃	CH₂F
148	CH ₃	2-CH ₃	CHF ₂
149	CH₃	2-CH ₃	CH₂CF₃
150	CH ₃	2-CH₃	CH ₂ CH=CH ₂
151	CH₃	2-CH₃	CH₂CH=CHCH₃
152	CH₃	2-CH₃	CH ₂ CH=C(CH ₃) ₂
153	CH₃	2-CH₃	ĈH₂CH=CHCI

Compound No.	R₂	(R ₅) _n	A-R ₇

154	CH₃	2-CH₃	CH ₂ CH=CCl ₂
155	CH₃	2-CH₃	CH ₂ C(CH ₃)=CH ₂
156	CH₃	2-CH₃ ,	CH₂C≡CH
157	CH ₃	2-CH ₃	CH₂Si(CH₃)₃
158	CH₃	2-CH ₃	CH ₂ -c.propyl-2,2-Cl ₂
159	CH₃	2-CH₃	CH₂CN
160	CH3	2-CH₃	CH₂COOC₂H₅
161	CH ₃	2-CH₃	CH(CH ₃)COOC ₂ H ₅
162	CH ₃	2-CH₃	CH₂C ₆ H₄-3-CF₃
163	CH ₃	2-CH₃	CH₂C ₆ H₄-4-F
164	CH₃	2-CH₃	CH₂C ₆ H₄-3-F
165	CH₃	2-CH₃	CH₂C ₆ H₄-2-F
166	CH₃	2-CH ₃	C(=O)OC ₂ H ₅
167	CH ₃	2-CH₃	C(=O)NHCH₃
168	CH3	2-CH₃	$C(=O)C(=O)OC_2H_5$
169	CH₃	4-OCH₂Si(CH₃)₃	CH₃
170	CH ₃	4-OCH ₂ Si(CH ₃) ₃	C ₂ H ₅
171	CH₃	4-OCH ₂ Si(CH ₃) ₃	n-C₃H ₇
172	CH ₃	4-OCH ₂ Si(CH ₃) ₃	i-C ₃ H ₇
173	CH₃	4-OCH ₂ Si(CH ₃) ₃	n-C ₄ H ₉
174	CH₃	4-OCH ₂ Si(CH ₃) ₃	n-C ₆ H ₁₃
175	CH₃	4-OCH ₂ Si(CH ₃) ₃	CH₂F
176	CH ₃	4-OCH₂Si(CH₃)₃	CHF ₂
177	CH₃	4-OCH ₂ Si(CH ₃) ₃	CH₂CF₃
178	CH ₃	4-OCH ₂ Si(CH ₃) ₃	CH ₂ CH=CH ₂
179	CH₃	4-OCH ₂ Si(CH ₃) ₃	CH₂CH=CHCH₃
180	CH₃	4-OCH ₂ Si(CH ₃) ₃	CH ₂ CH=C(CH ₃) ₂
181	CH₃	4-OCH ₂ Si(CH ₃) ₃	CH₂CH=CHCl
182	CH₃	4-OCH ₂ Si(CH ₃) ₃	CH ₂ CH=CCl ₂
183.	CH₃	4-OCH ₂ Si(CH ₃) ₃	[™] ČH₂C(CH₃)=CH₂

Compound	R ₂	· (R₅) _n	A-R ₇
No.		2	
184	СН₃	4-OCH₂Si(CH₃)₃	CH₂C≅CH
185	CH₃	4-OCH ₂ Si(CH ₃) ₃	CH₂Si(CH₃)₃
186	СН₃	4-OCH ₂ Si(CH ₃) ₃	CH ₂ -c.propyl-2,2-Cl ₂
187	CH₃	4-OCH₂Si(CH₃)₃	CH₂CN
188	CH₃	4-OCH ₂ Si(CH ₃) ₃	CH ₂ COOC ₂ H ₅
189	CH₃	4-OCH ₂ Si(CH ₃) ₃	CH(CH ₃)COOC ₂ H ₅
190	CH₃	4-OCH ₂ Si(CH ₃) ₃	CH ₂ C ₆ H ₄ -3-CF ₃
191	CH ₃	4-OCH ₂ Si(CH ₃) ₃	CH ₂ C ₆ H ₄ -4-F
192	СН₃	4-OCH ₂ Si(CH ₃) ₃	CH ₂ C ₆ H ₄ -3-F
193	CH₃	4-OCH₂Si(CH₃)₃	CH₂C ₆ H₄-2-F
194	CH ₃	4-OCH₂Si(CH₃)₃	C(=O)OC ₂ H ₅
195	CH₃	4-OCH₂Si(CH₃)₃	C(=O)NHCH ₃
196	CH₃	4-OCH ₂ Si(CH ₃) ₃	C(=O)C(=O)OC ₂ H ₅
197	CH₃	4-OCH ₂ C ₆ H ₄ -4-CF ₃	CH₃
198	CH₃	4-OCH ₂ C ₆ H ₄ -4-CF ₃	C₂H₅
199	CH₃	4-OCH ₂ C ₆ H ₄ -4-CF ₃	n-C ₃ H ₇
200	СН₃	4-OCH ₂ C ₆ H ₄ -4-CF ₃	i-C ₃ H ₇
201	CH₃	4-OCH ₂ C ₆ H ₄ -4-CF ₃	n-C₄H₀
202	CH ₃	4-OCH ₂ C ₆ H ₄ -4-CF ₃	n-C ₆ H ₁₃
203	CH₃	4-OCH ₂ C ₆ H ₄ -4-CF ₃	CH₂F
204	СН₃	4-OCH ₂ C ₆ H ₄ -4-CF ₃	CHF₂
205	CH₃	4-OCH ₂ C ₆ H ₄ -4-CF ₃	CH ₂ CF ₃
206	CH ₃	4-OCH ₂ C ₆ H ₄ -4-CF ₃	CH ₂ CH=CH ₂
207	CH₃	4-OCH ₂ C ₆ H ₄ -4-CF ₃	CH₂CH=CHCH₃
208	CH₃	4-OCH ₂ C ₆ H ₄ -4-CF ₃	CH ₂ CH=C(CH ₃) ₂
209	CH ₃	4-OCH ₂ C ₆ H ₄ -4-CF ₃	CH₂CH=CHCI
210	CH₃	4-OCH ₂ C ₆ H ₄ -4-CF ₃	CH2CH=CCI2
211	CH₃	4-OCH ₂ C ₆ H ₄ -4-CF ₃	CH ₂ C(CH ₃)=CH ₂
212	CH₃	4-OCH ₂ C ₆ H ₄ -4-CF ₃	CH₂C≡CH
213	CH ₃	4-OCH ₂ C ₆ H ₄ -4-CF ₃	ĈH₂Si(CH₃)₃

Compound	R ₂	(R ₅) _n	A-R ₇
No.	2		
214	CH₃	4-OCH₂C ₆ H₄-4-CF₃	CH ₂ -c.propyl-2,2-Cl ₂
215	CH₃	4-OCH ₂ C ₆ H ₄ -4-CF ₃	CH₂CN
216	CH₃	4-OCH ₂ C ₆ H ₄ -4-CF ₃	CH2COOC2H5
217	CH₃	4-OCH ₂ C ₆ H ₄ -4-CF ₃	CH(CH₃)COOC₂H₅
218	CH₃	4-OCH ₂ C ₆ H ₄ -4-CF ₃	CH ₂ C ₆ H ₄ -3-CF ₃
219	CH₃	4-OCH ₂ C ₆ H ₄ -4-CF ₃	CH ₂ C ₆ H ₄ -4-F
220	CH₃	4-OCH ₂ C ₆ H ₄ -4-CF ₃	CH ₂ C ₆ H ₄ -3-F
221	CH₃	4-OCH ₂ C ₆ H ₄ -4-CF ₃	CH ₂ C ₆ H ₄ -2-F
222	CH₃	4-OCH ₂ C ₆ H ₄ -4-CF ₃	C(=O)OC ₂ H ₅
223	CH₃	4-OCH₂C₅H₄-4-CF₃	C(=O)NHCH₃
224	CH₃	4-OCH ₂ C ₆ H ₄ -4-CF ₃	C(=O)C(=O)OC ₂ H ₅
225	CH₃	4-OCH ₂ C ₆ H ₄ -3-CF ₃	CH ₃
226	CH₃	4-OCH ₂ C ₆ H ₄ -3-CF ₃	C₂H₅
227	CH₃	4-OCH ₂ C ₆ H ₄ -3-CF ₃	n-C ₃ H ₇
228	CH ₃	4-OCH ₂ C ₆ H ₄ -3-CF ₃	i-C ₃ H ₇
229	CH₃	4-OCH ₂ C ₆ H ₄ -3-CF ₃	n-C ₄ H ₉
230	CH₃	4-OCH ₂ C ₆ H ₄ -3-CF ₃	n-C ₆ H ₁₃
231	CH₃	4-OCH ₂ C ₆ H ₄ -3-CF ₃	CH₂F
232	CH ₃	4-OCH ₂ C ₆ H ₄ -3-CF ₃	CHF ₂
233	CH ₃	4-OCH ₂ C ₆ H ₄ -3-CF ₃	CH₂CF₃
234	CH₃	4-OCH ₂ C ₆ H ₄ -3-CF ₃	CH ₂ CH=CH ₂
235	CH₃	4-OCH ₂ C ₆ H ₄ -3-CF ₃	CH₂CH=CHCH₃
236	CH₃	4-OCH ₂ C ₆ H ₄ -3-CF ₃	CH ₂ CH=C(CH ₃) ₂
237	CH₃	4-OCH ₂ C ₆ H ₄ -3-CF ₃	CH₂CH=CHCI
238	CH ₃	4-OCH ₂ C ₆ H ₄ -3-CF ₃	CH2CH=CCI2
239	CH ₃	4-OCH ₂ C ₆ H ₄ -3-CF ₃	CH ₂ C(CH ₃)=CH ₂
240	CH ₃	4-OCH ₂ C ₆ H ₄ -3-CF ₃	CH ₂ C≡CH
241	CH₃	4-OCH ₂ C ₆ H ₄ -3-CF ₃	CH₂Si(CH₃)₃
242	CH₃	4-OCH₂C6H₄-3-CF3	CH₂-c.propyl-2,2-Cl₂
243	CH ₃	4-OCH ₂ C ₆ H ₄ -3-CF ₃	ČH₂CN

Compound	R ₂	(R ₅) _n	A-R ₇
No.			
244	CH ₃	4-OCH ₂ C ₆ H ₄ -3-CF ₃	CH₂COOC₂H₅
245	CH ₃	4-OCH ₂ C ₆ H ₄ -3-CF ₃	CH(CH ₃)COOC ₂ H ₅
246	CH ₃	4-OCH ₂ C ₆ H ₄ -3-CF ₃	CH ₂ C ₆ H ₄ -3-CF ₃
247	CH ₃	4-OCH ₂ C ₆ H ₄ -3-CF ₃	CH ₂ C ₆ H ₄ -4-F
248	CH ₃	4-OCH ₂ C ₆ H ₄ -3-CF ₃	CH ₂ C ₆ H ₄ -3-F
249	CH₃	4-OCH ₂ C ₆ H ₄ -3-CF ₃	CH ₂ C ₆ H ₄ -2-F
250	CH₃	4-OCH ₂ C ₆ H ₄ -3-CF ₃	$C(=O)OC_2H_5$
251	CH₃	4-OCH ₂ C ₆ H ₄ -3-CF ₃	C(=O)NHCH ₃
252	CH₃	4-OCH ₂ C ₆ H ₄ -3-CF ₃	$C(=O)C(=O)OC_2H_5$
253	CH₃	4-OCH ₂ C ₆ H ₄ -2-CF ₃	CH₃
254	CH₃	4-OCH ₂ C ₆ H ₄ -2-CF ₃	C₂H₅
255	CH₃	4-OCH ₂ C ₆ H ₄ -2-CF ₃	n-C ₃ H ₇
256	CH₃	4-OCH ₂ C ₆ H ₄ -2-CF ₃	i-C ₃ H ₇
257	CH₃	4-OCH ₂ C ₆ H ₄ -2-CF ₃	n-C ₄ H ₉
258	CH ₃	4-OCH ₂ C ₆ H ₄ -2-CF ₃	n-C ₆ H ₁₃
259	CH ₃	4-OCH ₂ C ₆ H ₄ -2-CF ₃	CH₂F
260	CH ₃	4-OCH₂C ₆ H₄-2-CF₃	CHF ₂
261	CH₃	4-OCH ₂ C ₆ H ₄ -2-CF ₃	CH₂CF₃
262	CH ₃	4-OCH₂C ₆ H₄-2-CF₃	CH₂CH=CH₂
263	CH₃	4-OCH ₂ C ₆ H ₄ -2-CF ₃	CH₂CH=CHCH₃
264	CH ₃	4-OCH ₂ C ₆ H ₄ -2-CF ₃	CH ₂ CH=C(CH ₃) ₂
265	CH ₃	4-OCH ₂ C ₆ H ₄ -2-CF ₃	CH₂CH=CHCI
266	CH ₃	4-OCH ₂ C ₆ H ₄ -2-CF ₃	CH ₂ CH=CCl ₂
267	CH₃	4-OCH ₂ C ₆ H ₄ -2-CF ₃	CH ₂ C(CH ₃)=CH ₂
268	CH ₃	4-OCH ₂ C ₆ H ₄ -2-CF ₃	CH₂C≅CH
269	CH₃	4-OCH ₂ C ₆ H ₄ -2-CF ₃	CH₂Si(CH₃)₃
270	CH₃	4-OCH ₂ C ₆ H ₄ -2-CF ₃	CH ₂ -c.propyl-2,2-Cl ₂
271	CH ₃	4-OCH ₂ C ₆ H ₄ -2-CF ₃	CH₂CN
272	CH₃	4-OCH ₂ C ₆ H ₄ -2-CF ₃	CH₂COOC₂H₅
273	CH₃	4-OCH ₂ C ₆ H ₄ -2-CF ₃	ÜH(CH₃)COOC₂H₅

Compound No.	R ₂	, (R _s) _n	A-R ₇
274	CH₃	4-OCH ₂ C ₆ H ₄ -2-CF ₃	CH₂C ₆ H₄-3-CF₃
275	CH ₃	4-OCH ₂ C ₆ H ₄ -2-CF ₃	CH ₂ C ₆ H ₄ -4-F
276	CH₃	4-OCH₂C₀H₄-2-CF₃	CH₂C ₆ H₄-3-F
277	CH₃	4-OCH ₂ C ₆ H ₄ -2-CF ₃	CH₂C ₆ H₄-2-F
278	CH₃	4-OCH ₂ C ₆ H ₄ -2-CF ₃	$C(=O)OC_2H_5$
279	CH₃	4-OCH₂C ₆ H₄-2-CF₃	C(=O)NHCH ₃
280	CH₃	4-OCH ₂ C ₆ H ₄ -2-CF ₃	C(=O)C(=O)OC ₂ H ₅
281	CH₃	4-OCH ₂ C ₆ H ₄ -4-F	CH ₃
282	CH₃	4-OCH ₂ C ₆ H ₄ -4-F	C ₂ H ₅
283	CH ₃	4-OCH ₂ C ₆ H ₄ -4-F	n-C₃H ₇
284	CH ₃	4-OCH ₂ C ₆ H ₄ -4-F	i-C₃H ₇
285	CH₃	4-OCH ₂ C ₆ H ₄ -4-F	n-C₄H ₉
286	CH₃	4-OCH ₂ C ₆ H ₄ -4-F	n-C ₆ H ₁₃
287	CH ₃	4-OCH ₂ C ₆ H ₄ -4-F	CH₂F
288	CH ₃	4-OCH ₂ C ₆ H ₄ -4-F	CHF ₂
289	CH₃	4-OCH ₂ C ₆ H ₄ -4-F	CH₂CF₃
290	CH3	4-OCH ₂ C ₆ H ₄ -4-F	CH₂CH=CH₂
291	CH₃	4-OCH ₂ C ₆ H ₄ -4-F	CH₂CH=CHCH₃
292	CH₃	4-OCH ₂ C ₆ H ₄ -4-F	CH ₂ CH=C(CH ₃) ₂
293	CH₃	4-OCH ₂ C ₆ H ₄ -4-F	CH₂CH=CHCI
294	CH₃	4-OCH ₂ C ₆ H ₄ -4-F	CH ₂ CH=CCl ₂
295	CH₃	4-OCH ₂ C ₆ H ₄ -4-F	$CH_2C(CH_3)=CH_2$
296	CH₃	4-OCH ₂ C ₆ H ₄ -4-F	CH₂C≅CH
297	CH ₃	4-OCH ₂ C ₆ H ₄ -4-F	CH₂Si(CH₃)₃
298	CH₃	4-OCH ₂ C ₆ H ₄ -4-F	CH ₂ -c.propyl-2,2-Cl ₂
299	CH₃	4-OCH ₂ C ₆ H ₄ -4-F	CH₂CN
300	CH3	4-OCH ₂ C ₆ H ₄ -4-F	CH ₂ COOC ₂ H ₅
301	CH₃	4-OCH ₂ C ₆ H ₄ -4-F	CH(CH₃)COOC₂H₅
302	CH₃	4-OCH ₂ C ₆ H ₄ -4-F	°CH₂C6H₄-3-CF₃
303	CH₃	4-OCH₂C6H4-4-F	CH₂C ₆ H₄-4-F

Compound	R ₂	(R _s) _n	A-R ₇
No.		>	
304	СН₃	4-OCH₂C₀H₄-4-F	CH₂C₀H₄-3-F
305	CH₃	4-OCH ₂ C ₆ H ₄ -4-F	CH₂C ₆ H₄-2-F
306	CH ₃	4-OCH ₂ C ₆ H ₄ -4-F	C(=O)OC ₂ H ₅
307	CH₃	4-OCH₂C ₆ H₄-4-F	C(=O)NHCH ₃
308	CH₃	4-OCH ₂ C ₆ H ₄ -4-F	$C(=O)C(=O)OC_2H_5$
309	CH₃	4-OC ₆ H ₄ -3-CF ₃	CH₃
310	CH₃	4-OC ₆ H ₄ -3-CF ₃	C ₂ H ₅
311	CH₃	4-OC ₆ H ₄ -3-CF ₃	n-C ₃ H ₇
312	CH3	4-OC ₆ H ₄ -3-CF ₃	i-C ₃ H ₇
313	CH₃	4-OC ₆ H ₄ -3-CF ₃	n-C₄H ₉
314	CH ₃	4-OC ₆ H ₄ -3-CF ₃	n-C ₆ H ₁₃
315	CH₃	4-OC ₆ H ₄ -3-CF ₃	CH₂F
316	CH₃	4-OC ₆ H ₄ -3-CF ₃	CHF ₂
317	CH₃	4-OC ₆ H ₄ -3-CF ₃	CH₂CF₃
318	CH₃	4-OC ₆ H ₄ -3-CF ₃	CH ₂ CH=CH ₂
319 .	CH ₃	4-OC ₆ H ₄ -3-CF ₃	CH₂CH=CHCH₃
320	CH ₃	4-OC ₆ H ₄ -3-CF ₃	CH ₂ CH=C(CH ₃) ₂
321	CH ₃	4-OC ₆ H ₄ -3-CF ₃	CH₂CH=CHCI
322	CH₃	4-OC ₆ H ₄ -3-CF ₃	CH ₂ CH=CCl ₂
323	CH₃	4-OC ₆ H ₄ -3-CF ₃	CH ₂ C(CH ₃)=CH ₂
324	CH ₃	4-OC ₆ H ₄ -3-CF ₃	CH₂C≡CH
325	CH₃	4-OC ₆ H ₄ -3-CF ₃	CH₂Si(CH₃)₃
326	CH₃	4-OC ₆ H ₄ -3-CF ₃	CH2-c.propyl-2,2-Cl2
327	CH₃	4-OC ₆ H ₄ -3-CF ₃	CH₂CN
328	CH ₃	4-OC ₆ H ₄ -3-CF ₃	CH₂COOC₂H₅
329	CH₃	4-OC ₆ H ₄ -3-CF ₃	CH(CH ₃)COOC ₂ H ₅
330	CH ₃	4-OC ₆ H ₄ -3-CF ₃	CH₂C ₆ H₄-3-CF₃
331	CH ₃	4-OC ₆ H ₄ -3-CF ₃	CH₂C ₆ H₄-4-F
332	CH₃	4-OC ₆ H ₄ -3-CF ₃	CH₂C ₆ H₄-3-F
333	CH₃	4-OC ₆ H ₄ -3-CF ₃	ČH₂C ₆ H₄-2-F

Compound No.	R₂	(R ₅) _n	A-R ₇
334	CH₃	4-OC ₆ H ₄ -3-CF ₃	C(=O)OC ₂ H ₅
335	CH₃	4-OC₀H₄-3-CF₃	C(=O)NHCH₃
336	CH₃	4-OC₅Ḥ₄-3-CF₃	C(=O)C(=O)OC ₂ H ₅
337	C ₂ H ₅	4-OCH₂C₀H₄-3-CF₃	CH₃
338	C₂H₅	4-OCH ₂ C ₆ H ₄ -3-CF ₃	C₂H₅
339	C₂H₅	4-OCH ₂ C ₆ H ₄ -3-CF ₃	n-C ₃ H ₇
340	C ₂ H ₅	4-OCH ₂ C ₆ H ₄ -3-CF ₃	i-C ₃ H ₇
341	C ₂ H ₅	4-OCH ₂ C ₆ H ₄ -3-CF ₃	n-C ₄ H ₉
342	C ₂ H ₅	4-OCH ₂ C ₆ H ₄ -3-CF ₃	n-C ₆ H ₁₃
343	C₂H₅	4-OCH ₂ C ₆ H ₄ -3-CF ₃	CH₂F
344	C ₂ H ₅	4-OCH ₂ C ₆ H ₄ -3-CF ₃	CHF ₂
345	C ₂ H ₅	4-OCH ₂ C ₆ H ₄ -3-CF ₃	CH₂CF₃
346	C ₂ H ₅	4-OCH₂C6H₄-3-CF₃	CH ₂ CH=CH ₂
347	C ₂ H ₅	4-OCH ₂ C ₆ H ₄ -3-CF ₃	CH₂CH=CHCH₃
348	C ₂ H ₅	4-OCH ₂ C ₆ H ₄ -3-CF ₃	CH ₂ CH=C(CH ₃) ₂
349	C₂H₅	4-OCH ₂ C ₆ H ₄ -3-CF ₃	CH₂CH=CHCI
350	C ₂ H ₅	4-OCH ₂ C ₆ H ₄ -3-CF ₃	CH₂CH=CCI₂
351	C ₂ H ₅	4-OCH ₂ C ₆ H ₄ -3-CF ₃	CH ₂ C(CH ₃)=CH ₂
352	C ₂ H ₅	4-OCH ₂ C ₆ H ₄ -3-CF ₃	CH₂C≒CH
353	C ₂ H ₅	4-OCH ₂ C ₆ H ₄ -3-CF ₃	CH₂Si(CH₃)₃
354	C₂H₅	4-OCH₂C6H4-3-CF3	CH ₂ -c.propyl-2,2-Cl ₂
355	C₂H₅	4-OCH ₂ C ₆ H ₄ -3-CF ₃	CH₂CN
356	C ₂ H ₅	4-OCH ₂ C ₆ H ₄ -3-CF ₃	CH₂COOC₂H₅
357	C ₂ H ₅	4-OCH ₂ C ₆ H ₄ -3-CF ₃	CH(CH₃)COOC₂H₅
358	C ₂ H ₅	4-OCH ₂ C ₆ H ₄ -3-CF ₃	CH ₂ C ₆ H ₄ -3-CF ₃
359	C ₂ H ₅	4-OCH ₂ C ₆ H ₄ -3-CF ₃	CH₂C ₆ H₄-4-F
360	C ₂ H ₅	4-OCH ₂ C ₆ H ₄ -3-CF ₃	CH₂C ₆ H₄-3-F
361	C ₂ H ₅	4-OCH ₂ C ₆ H ₄ -3-CF ₃	CH ₂ C ₆ H ₄ -2-F
362	C ₂ H ₅	4-OC ₆ H ₄ -3-CF ₃	C(=O)OC ₂ H ₅
363	C ₂ H ₅	4-OCH ₂ C ₆ H ₄ -3-CF ₃	ີັC(=O)NHCH₃

Compound	R ₂	(R ₅) _n	A-R ₇
No.		· · · · · · · · · · · · · · · · · · ·	;
364	C₂H₅	4-OCH₂C₀H₄-3-CF₃	C(=O)C(=O)OC ₂ H ₅
365	СН₃	4-OC ₆ H ₄ -4-CI	CH₃
366	CH ₃	4-OC ₆ H ₄ -4-Cl	C₂H₅
367	CH ₃	4-OC ₆ H₄-4-CI	n-C ₃ H ₇
368	CH₃	4-OC ₆ H ₄ -4-Cl	i-C₃H ₇
369	CH ₃	4-OC ₆ H ₄ -4-Cl	n-C₄H ₉
370	CH₃	4-OC ₆ H ₄ -4-Cl	n-C ₆ H ₁₃
371	CH ₃	4-OC ₆ H ₄ -4-Cl	CH₂F
372	CH₃	4-OC ₆ H ₄ -4-Cl	CHF ₂
373	CH ₃	4-OC ₆ H ₄ -4-Cl	CH₂CF₃
374	CH ₃	4-OC ₆ H ₄ -4-Cl	CH₂CH=CH₂
375	CH₃	4-OC ₆ H ₄ -4-Cl	CH₂CH=CHCH₃
376	CH₃	4-OC ₆ H ₄ -4-Cl	CH ₂ CH=C(CH ₃) ₂
377	CH ₃	4-OC ₆ H ₄ -4-Cl	CH₂CH=CHCI
378	CH ₃	4-OC ₆ H ₄ -4-Cl	CH ₂ CH=CCI ₂
379	CH₃	4-OC ₆ H ₄ -4-Cl	CH ₂ C(CH ₃)=CH ₂
380	CH ₃	4-OC ₆ H ₄ -4-Cl	CH₂C≡CH
381	CH₃	4-OC ₆ H ₄ -4-Cl	CH ₂ Si(CH ₃) ₃
382	CH₃	4-OC ₆ H ₄ -4-Cl	CH ₂ -c.propyl-2,2-Cl ₂
383	CH₃	4-OC ₆ H ₄ -4-CI	CH₂CŅ
384	CH₃	4-OC ₆ H ₄ -4-CI	CH₂COOC₂H₅
385	CH ₃	4-OC ₆ H ₄ -4-Cl	CH(CH₃)COOC₂H₅
386	CH ₃	4-OC ₆ H ₄ -4-Cl	CH₂C ₆ H₄-3-CF₃
387	CH₃	4-OC ₆ H ₄ -4-Cl	CH₂C ₆ H₄-4-F
388	CH ₃	4-OC ₆ H ₄ -4-Cl	CH ₂ C ₆ H ₄ -3-F
389	CH₃	4-OC ₆ H ₄ -4-Cl	CH₂C ₆ H₄-2-F
390	CH₃	4-OC ₆ H ₄ -4-Cl	$C(=O)OC_2H_5$
391	CH₃	4-OC ₆ H ₄ -4-Cl	C(=O)NHCH ₃
392	CH₃	4-OC ₆ H ₄ -4-Cl	$C(=O)C(=O)OC_2H_5$
393	CH₃	4-OC ₆ H ₄ -3-CI	ĈH₃

Compound	R ₂	(R ₅) _n	A-R ₇
No.	3		
394	СН₃	4-OC₅H₄-3-CI	C₂H₅
395	CH₃	4-OC ₆ H ₄ -3-Cl	n-C₃H ₇
396	CH₃	4-OC ₆ H ₄ -3-CI	i-C₃H ₇
397	CH₃	4-OC ₆ H ₄ -3-CI	n-C₄H ₉
398	CH₃	4-OC ₆ H ₄ -3-CI	n-C ₆ H ₁₃
399	CH₃	4-OC ₆ H ₄ -3-Cl	CH₂F
400	CH₃	4-OC ₆ H ₄ -3-Cl	CHF ₂
401	CH ₃	4-OC ₆ H ₄ -3-Cl	CH₂CF₃
402	CH₃	4-OC ₆ H ₄ -3-CI	CH₂CH=CH₂
403	CH₃	4-OC ₆ H ₄ -3-Cl	CH₂CH=CHCH₃
404	CH₃	4-OC ₆ H ₄ -3-Cl	CH ₂ CH=C(CH ₃) ₂
405	CH₃	4-OC ₆ H ₄ -3-Cl	CH₂CH=CHCI
406	CH₃	4-OC ₆ H ₄ -3-CI	CH2CH=CCI2
407	CH₃	4-OC ₆ H ₄ -3-Cl	CH ₂ C(CH ₃)=CH ₂
408	CH ₃	4-OC ₆ H ₄ -3-CI	CH ₂ C≡CH
409	CH₃	4-OC ₆ H ₄ -3-CI	CH₂Si(CH₃)₃
410	CH₃	4-OC ₆ H ₄ -3-Cl	CH ₂ -c.propyl-2,2-Cl ₂
411	CH₃	4-OC ₆ H ₄ -3-CI	CH₂CN
412	CH₃	4-OC ₆ H ₄ -3-CI	CH₂COOC₂H₅
413	CH₃	4-OC ₆ H ₄ -3-Cl	CH(CH ₃)COOC ₂ H ₅
414	CH₃	4-OC ₆ H ₄ -3-Cl	CH₂C ₆ H₄-3-CF₃
415	CH₃	4-OC ₆ H ₄ -3-Cl	CH ₂ C ₆ H ₄ -4-F
416	CH₃	4-OC ₆ H ₄ -3-Cl	CH₂C ₆ H₄-3-F
417	CH₃	4-OC ₆ H ₄ -3-Cl	CH₂C ₆ H₄-2-F
418	СН₃	4-OC ₆ H ₄ -3-Cl	C(=O)OC ₂ H ₅
419	CH₃	4-OC ₆ H ₄ -3-Cl	C(=O)NHCH₃
420	CH₃	4-OC ₆ H ₄ -3-Cl	C(=0)C(=0)OC ₂ H ₅
421	CH ₃	4-OC ₆ H ₄ -2-Cl	CH₃
242	CH ₃	4-OC ₆ H ₄ -2-Cl	C₂H₅
423	CH₃	4-OC ₆ H ₄ -2-Cl	n-C ₃ H ₇

Compound	R ₂	(R _s) _n	A-R ₇
No.	2		
424	CH₃	4-0C ₆ H ₄ -2-Cl	i-C ₃ H ₇
425	CH₃	4-OC ₆ H ₄ -2-Cl	n-C₄H ₉
426	CH ₃	4-OC6H4-2-CI	n-C ₆ H ₁₃
427	CH₃	4-OC ₆ H ₄ -2-Cl	CH₂F
428	CH₃	4-OC ₆ H ₄ -2-Cl	CHF ₂
429	CH ₃	4-OC6H4-2-CI	CH₂CF₃
430	CH ₃	4-OC5H4-2-CI	CH ₂ CH=CH ₂
431	CH ₃	4-OC ₆ H ₄ -2-Cl	CH₂CH=CHCH₃
432	CH₃	4-OC ₆ H ₄ -2-Cl	CH ₂ CH=C(CH ₃) ₂
433	CH ₃	4-OC ₆ H ₄ -2-Cl	CH₂CH=CHCI
434	CH ₃	4-OC ₆ H ₄ -2-Cl	CH ₂ CH=CCl ₂
435	CH ₃	4-OC ₆ H₄-2-CI	CH ₂ C(CH ₃)=CH ₂
436	CH₃	4-OC ₆ H ₄ -2-Cl	CH₂C≡CH
437	CH₃	4-OC ₆ H ₄ -2-Cl	CH₂Si(CH₃)₃
438	CH₃	4-OC ₆ H ₄ -2-Cl	CH ₂ -c.propyl-2,2-Cl ₂
439	CH₃	4-OC ₆ H ₄ -2-Cl	CH₂CN
440	CH₃	4-OC ₆ H ₄ -2-Cl	CH ₂ COOC ₂ H ₅
441	CH₃	4-OC ₆ H ₄ -2-Cl	CH(CH₃)COOC₂H₅
442	CH ₃	4-OC ₆ H ₄ -2-Cl	CH ₂ C ₆ H ₄ -3-CF ₃
443	CH₃	4-OC ₆ H ₄ -2-Cl	CH₂C ₆ H₄-4-F
444	CH ₃	4-OC ₆ H ₄ -2-Cl	CH₂C ₆ H₄-3-F
445	CH₃	4-OC ₆ H ₄ -2-Cl	CH₂C ₆ H₄-2-F
446	CH ₃	4-OC ₆ H ₄ -2-CI	$C(=O)OC_2H_5$
447	CH₃	4-OC ₆ H₄-2-CI	C(=O)NHCH₃
448	CH₃	4-OC ₆ H ₄ -2-Cl	$C(=O)C(=O)OC_2H_5$
449	CH₃	4-OC ₆ H ₄ -4-F	CH₃
450	CH₃	4-OC ₆ H ₄ -4-F	C₂H₅
451	CH₃	4-OC ₆ H ₄ -4-F	n-C ₃ H ₇
452	CH₃	4-OC ₆ H ₄ -4-F	i-C ₃ H ₇
453	СН₃	4-OC ₆ H ₄ -4-F	ີ້n-C₄H ₉

Compound	R ₂	(R ₅) _n	A-R ₇
No.		2	
454	CH₃	4-OC₀H₄-4-F	n-C ₆ H ₁₃ .
455	CH ₃	4-OC ₆ H ₄ -4-F	CH₂F
456	CH₃	4-OC ₆ H₄-4-F	CHF ₂
457	ÇH₃	4-OC ₆ H ₄ -4-F	CH₂CF₃
458	CH ₃	4-0C ₆ H ₄ -4-F	CH ₂ CH=CH ₂
459	CH ₃	4-OC ₆ H ₄ -4-F	CH₂CH=CHCH₃
460	CH₃	4-0C ₆ H ₄ -4-F	CH ₂ CH=C(CH ₃) ₂
461	CH₃	4-0C ₆ H ₄ -4-F	CH₂CH=CHCI
462	CH₃	4-OC ₆ H ₄ -4-F	CH ₂ CH=CCl ₂
463	СН₃	4-OC ₆ H ₄ -4-F	CH ₂ C(CH ₃)=CH ₂
464	CH₃	4-OC ₆ H ₄ -4-F	CH₂C≡CH
465	CH₃	4-0C ₆ H ₄ -4-F	CH ₂ Si(CH ₃) ₃
466	CH₃	4-OC ₆ H ₄ -4-F	CH ₂ -c.propyl-2,2-Cl ₂
467	CH₃	4-0C ₆ H ₄ -4-F	CH₂CN
468	CH ₃	4-OC ₆ H ₄ -4-F	CH ₂ COOC ₂ H ₅
469	CH₃	4-OC ₆ H ₄ -4-F	CH(CH ₃)COOC ₂ H ₅
470	CH ₃	4-OC ₆ H ₄ -4-F	CH ₂ C ₆ H ₄ -3-CF ₃
471	CH ₃	4-OC ₆ H ₄ -4-F	CH ₂ C ₆ H ₄ -4-F
472	CH₃	4-OC ₆ H ₄ -4-F	CH₂C ₆ H₄-3-F
473	CH₃	4-OC ₆ H ₄ -4-F	CH₂C ₆ H₄-2-F
474	CH₃	4-OC ₆ H ₄ -4-F	C(=O)OC ₂ H ₅
475	CH ₃	4-OC ₆ H ₄ -4-F	C(=O)NHCH ₃
476	CH₃	4-OC ₆ H ₄ -4-F	$C(=O)C(=O)OC_2H_5$
477	CH ₃	4-OC ₆ H ₄ -3-F	CH₃
478	CH₃	4-OC ₆ H ₄ -3-F	C₂H₅
479	CH₃	4-OC ₆ H ₄ -3-F	n-C₃H ₇
480	CH₃	4-OC ₆ H ₄ -3-F	i-C ₃ H ₇
481	CH₃	4-OC ₆ H ₄ -3-F	n-C₄H ₉
482	CH₃	4-OC ₆ H ₄ -3-F	n-C ₆ H ₁₃
483	CH₃	4-OC ₆ H ₄ -3-F	CH₂F

Compound	R₂	(R₅)n	A-R ₇
No.	,		
484	CH₃	4-OC ₆ H₄-3-F	CHF₂
485	CH₃	4-OC ₆ H ₄ -3-F	CH₂CF₃
486	CH₃	4-OC₅H₄-3-F	CH₂CH=CH₂
487	CH₃	4-OC ₆ H ₄ -3-F	CH₂CH=CHCH₃
488	CH₃	4-OC ₆ H ₄ -3-F	CH ₂ CH=C(CH ₃) ₂
489	CH₃	4-OC ₆ H ₄ -3-F	CH₂CH=CHCI
490	CH ₃	4-OC ₆ H ₄ -3-F	CH ₂ CH=CCl ₂
491	CH₃	4-OC ₆ H ₄ -3-F	CH ₂ C(CH ₃)=CH ₂
492	CH₃	4-OC ₆ H ₄ -3-F	CH₂C≘CH
493	CH₃	4-OC ₆ H ₄ -3-F	CH₂Si(CH₃)₃
494	CH ₃	4-OC ₆ H ₄ -3-F	CH ₂ -c.propyl-2,2-Cl ₂
495	CH₃	4-OC ₆ H ₄ -3-F	CH₂CN
496	CH₃	4-OC ₆ H ₄ -3-F	CH ₂ COOC ₂ H ₅
497	CH₃	4-OC ₆ H ₄ -3-F	CH(CH ₃)COOC ₂ H ₅
498	CH₃	4-OC ₆ H ₄ -3-F	CH ₂ C ₆ H ₄ -3-CF ₃
499	CH₃	4-OC ₆ H ₄ -3-F	CH ₂ C ₆ H ₄ -4-F
500	CH ₃	4-OC ₆ H ₄ -3-F	CH₂C ₆ H₄-3-F
501	CH₃	4-OC ₆ H ₄ -3-F	CH₂C ₆ H₄-2-F
502	CH₃	4-OC ₆ H ₄ -3-F	C(=O)OC ₂ H ₅
503	CH₃	4-OC ₆ H ₄ -3-F	C(=O)NHCH ₃
504	CH₃	4-OC ₆ H ₄ -3-F	$C(=0)C(=0)OC_2H_5$
505	CH₃	4-OC ₆ H ₄ -2-F	CH₃
506	CH₃	4-OC ₆ H ₄ -2-F	C ₂ H ₅
507	CH₃	4-OC ₆ H ₄ -2-F	n-C ₃ H ₇
508	СН₃	4-OC ₆ H ₄ -2-F	i-C ₃ H ₇
509	CH₃	4-OC ₆ H ₄ -2-F	n-C₄H ₉
510	CH₃	4-OC ₆ H ₄ -2-F	n-C ₆ H ₁₃
511	CH₃	4-OC ₆ H ₄ -2-F	CH₂F
512	CH₃	4-OC ₆ H ₄ -2-F	CHF ₂
513	CH₃	4-OC₀H₄-2-F	ີCH₂CF₃
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Compound	R₂	(R₅) _n	A-R ₇
No.	-	•	
514	СН₃	4-OC₀H₄-2-F	CH₂CH=CH₂
515	CH ₃	4-0C ₆ H ₄ -2-F	CH₂CH=CHCH₃
516	CH₃	4-OC ₆ H ₄ -2-F	CH ₂ CH=C(CH ₃) ₂
517	CH ₃	4-0C ₆ H ₄ -2-F	CH₂CH=CHCI
518	CH ₃	4-OC ₆ H ₄ -2-F	CH ₂ CH=CCl ₂
519	CH₃	4-OC ₆ H ₄ -2-F	CH ₂ C(CH ₃)=CH ₂
520	CH₃	4-OC ₆ H ₄ -2-F	CH₂C≡CH
521	CH ₃	4-OC ₆ H ₄ -2-F	CH ₂ Si(CH ₃) ₃
522	CH₃	4-OC ₆ H ₄ -2-F	CH ₂ -c.propyl-2,2-Cl ₂
523	CH₃	4-OC ₆ H ₄ -2-F	CH₂CN .
524	CH₃	4-OC ₆ H ₄ -2-F	CH ₂ COOC ₂ H ₅
525	CH₃	4-OC ₆ H ₄ -2-F	CH(CH₃)COOC₂H₅
526	CH₃	4-OC ₆ H ₄ -2-F	CH ₂ C ₆ H ₄ -3-CF ₃
527	CH₃	4-OC ₆ H ₄ -2-F	CH ₂ C ₆ H ₄ -4-F
528	CH₃	4-OC ₆ H ₄ -2-F	CH ₂ C ₆ H ₄ -3-F
529	CH ₃	4-OC ₆ H ₄ -2-F	CH₂C ₆ H₄-2-F
530	CH₃	4-OC ₆ H ₄ -2-F	$C(=O)OC_2H_5$
531	CH ₃	4-OC ₆ H ₄ -2-F	C(=O)NHCH₃
532	CH ₃	4-OC ₆ H ₄ -2-F	$C(=O)C(=O)OC_2H_5$
533	CH₃	4-OC ₆ H ₄ -4-Br	CH₃
534	CH ₃	4-OC ₆ H ₄ -4-Br	C ₂ H ₅
535	CH₃	4-OC ₆ H ₄ -4-Br	n-C ₃ H ₇
536	CH ₃	4-OC ₆ H ₄ -4-Br	i-C₃H ₇
537	CH₃	4-OC ₆ H ₄ -4-Br	n-C₄H ₉
538	CH₃	4-OC ₆ H ₄ -4-Br	n-C ₆ H ₁₃
539	CH ₃	4-OC ₆ H ₄ -4-Br	ÇH₂F
540	CH₃	4-OC ₆ H ₄ -4-Br	CHF₂
541	CH₃	4-OC ₆ H ₄ -4-Br	CH₂CF₃
542	CH₃	4-OC ₆ H ₄ -4-Br	CH₂CH=CH₂
543	CH ₃	4-OC ₆ H ₄ -4-Br	ິັCH₂CH=CHCH₃

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Compound	R ₂	(R ₅) _n	A-R ₇
No.		2	
544	CH₃	4-OC ₆ H ₄ -4-Br	CH₂CH=C(CH₃)₂
545	CH₃	4-OC ₆ H ₄ -4-Br	CH₂CH=CHCI
546	CH₃	4-OC ₆ H ₄ -4-Br	CH ₂ CH=CCI ₂
547	CH₃	4-OC₅H₄-4-Br	CH ₂ C(CH ₃)=CH ₂
548	CH₃	4-OC₀H₄-4-Br	CḤ₂C≡CH
549	CH₃	4-OC ₆ H ₄ -4-Br	CH ₂ Si(CH ₃) ₃
550	CH ₃	4-OC ₆ H ₄ -4-Br	CH ₂ -c.propyl-2,2-Cl ₂
551	CH₃	4-OC ₆ H ₄ -4-Br	CH₂CN
552	CH₃	4-OC ₆ H ₄ -4-Br	CH2COOC2H5
553	CH ₃	4-OC ₆ H ₄ -4-Br	CH(CH₃)COOC₂H₅
554	CH₃	4-OC ₆ H ₄ -4-Br	CH ₂ C ₆ H ₄ -3-CF ₃
555	CH₃	4-OC ₆ H ₄ -4-Br	CH₂C ₆ H₄-4-F
556	CH3	4-OC ₆ H ₄ -4-Br	CH₂C ₆ H₄-3-F
557	CH₃	4-OC ₆ H ₄ -4-Br	CH₂C ₆ H₄-2-F
558	CH ₃	4-OC ₆ H ₄ -4-Br	C(=O)OC ₂ H ₅
559	CH₃	4-OC ₆ H ₄ -4-Br	C(=O)NHCH ₃
560	CH₃	4-OC ₆ H₄-4-Br	C(=O)C(=O)OC ₂ H ₅
561	CH ₃	4-OC ₆ H ₄ -3-Br	CH₃
562	CH ₃	4-OC ₆ H₄-3-Br	C₂H₅
563	CH ₃	4-OC ₆ H ₄ -3-Br	n-C ₃ H ₇
564	CH₃	4-OC ₆ H ₄ -3-Br	i-C₃H ₇
565	CH₃	4-OC ₆ H ₄ -3-Br	n-C₄H ₉
566	CH₃	4-OC ₆ H₄-3-Br	n-C ₆ H ₁₃
567	CH₃	4-OC ₆ H ₄ -3-Br	CH₂F
568	CH₃	4-OC₅H₄-3-Br	CHF ₂
569	CH₃	4-OC₅H₄-3-Br	CH₂CF₃
570	CH₃	4-OC ₆ H ₄ -3-Br	CH ₂ CH=CH ₂
571	CH₃	4-OC ₆ H ₄ -3-Br	CH₂CH=CHCH₃
572	CH ₃	4-OC ₆ H ₄ -3-Br	CH₂CH=C(CH₃)₂
573	CH ₃	4-OC ₆ H₄-3-Br	CH₂CH=CHCI

Compound	R ₂	(R _s) _n	A-R ₇
No.		*	
574	СН₃	4-OC₀H₄-3-Br	CH₂CH=CCI₂
575	CH ₃	4-OC ₆ H ₄ -3-Br	CH ₂ C(CH ₃)=CH ₂
576	CH₃	4-OC₅H₄-3-Br	CH₂C≡CH
577	CH₃	4-OC₀H₄-3-Br	CH ₂ Si(CH ₃) ₃
578	CH₃	4-OC ₆ H ₄ -3-Br	CH ₂ -c.propyl-2,2-Cl ₂
579	CH₃	4-OC ₆ H ₄ -3-Br	CH₂CN
580	CH₃	4-0Ç ₆ H ₄ -3-Br	CH ₂ COOC ₂ H ₅
581	CH₃	4-OC ₆ H ₄ -3-Br	CH(CH ₃)COOC ₂ H ₅
582	CH₃	4-OC ₆ H ₄ -3-Br	CH ₂ C ₆ H ₄ -3-CF ₃
583	CH ₃	4-OC ₆ H ₄ -3-Br	CH ₂ C ₆ H ₄ -4-F
584	CH ₃	4-OC ₆ H ₄ -3-Br	CH₂C ₆ H₄-3-F
585	CH ₃	4-OC ₆ H ₄ -3-Br	CH₂C ₆ H₄-2-F
586	CH₃	4-OC ₆ H ₄ -3-Br	$C(=O)OC_2H_5$
587	CH₃	4-OC ₆ H ₄ -3-Br	C(=O)NHCH₃
588	CH₃	4-OC ₆ H ₄ -3-Br	$C(=O)C(=O)OC_2H_5$
589	CH₃	4-OC ₆ H ₄ -2-Br	CH₃
590	CH ₃	4-OC ₆ H ₄ -2-Br	C ₂ H ₅
591	CH ₃	4-OC ₆ H ₄ -2-Br	n-C₃H ₇
592	CH₃	4-OC ₆ H ₄ -2-Br	i-C₃H ₇
593	CH3	4-OÇ₅H₄-2-Br	n-C₄H ₉
594	CH₃	4-OC ₆ H₄-2-Br	n-C ₆ H ₁₃
595	CH₃	4-OC ₆ H₄-2-Br	CH₂F
596	CH₃	4-OC ₆ H ₄ -2-Br	CHF₂
597	CH ₃	4-OC ₆ H ₄ -2-Br	CH₂CF₃
598	CH₃	4-OC ₆ H₄-2-Br	CH₂CH=CH₂
599	CH ₃	4-OC ₆ H₄-2-Br	CH₂CH=CHCH₃
600	CH₃	4-OC ₆ H ₄ -2-Br	CH ₂ CH=C(CH ₃) ₂
601	CH₃	4-OC ₆ H ₄ -2-Br	CH₂CH=CHCI
602	CH₃	4-OC ₆ H ₄ -2-Br	.∞GH ₂ CH=CCl ₂
603	CH₃	4-OC ₆ H₄-2-Br	CH ₂ C(CH ₃)=CH ₂

Compound	d R₂	(R₅)n	A-R ₇
No.		2	
604	CH₃	4-OC₀H₄-2-Br	CH₂C≡CH
605	CH₃	4-OC ₆ H ₄ -2-Br	CH₂Si(CH₃)₃
606	CH₃	4-OC ₆ H ₄ -2-Br	CH ₂ -c.propyl-2,2-Cl ₂
607	CH₃	4-OC ₆ H ₄ -2-Br	CH₂CN
608	CH₃	4-OC ₆ H ₄ -2-Br	CH ₂ COOC ₂ H ₅
609	CH₃	4-OC ₆ H ₄ -2-Br	CH(CH ₃)COOC ₂ H ₅
610	CH₃	4-OÇ ₆ H₄-2-Br	CH ₂ C ₆ H ₄ -3-CF ₃
611	CH₃	4-OÇ₅H₄-2-Br	CH₂C ₆ H₄-4-F
612	CH₃	4-OC ₆ H ₄ -2-Br	CH₂C ₆ H₄-3-F
613	CH₃	4-OC ₆ H ₄ -2-Br	CH₂C ₆ H₄-2-F
614	CH₃	4-OC ₆ H ₄ -2-Br	$C(=O)OC_2H_5$
615	CH ₃	4-OC ₆ H ₄ -2-Br	C(=O)NHCH ₃
616	CH₃	4-OC ₆ H ₄ -2-Br	$C(=O)C(=O)OC_2H_5$
617	CH ₃	4-OC ₆ H ₃ -2,4-Cl ₂	CH₃
618	CH₃	4-OC ₆ H ₃ -2,4-Cl ₂	C ₂ H ₅
619	CH₃	4-OC ₆ H ₃ -2,4-Cl ₂	n-C₃H ₇
620	CH₃	4-OC ₆ H ₃ -2,4-Cl ₂	i-C ₃ H ₇
621	CH₃	4-OC ₆ H ₃ -2,4-Cl ₂	n-C ₄ H ₉
622	CH₃	4-OC ₆ H ₃ -2,4-Cl ₂	n-C ₆ H ₁₃
623	CH₃	4-OC ₆ H ₃ -2,4-Cl ₂	CH₂F
624	CH₃	4-OC ₆ H ₃ -2,4-Cl ₂	CHF₂
625	CH₃	4-OC ₆ H ₃ -2,4-Cl ₂	CH₂CF₃
626	CH₃	4-OC ₆ H ₃ -2,4-Cl ₂	CH₂CH=CH₂
627	CH₃	4-OC ₆ H ₃ -2,4-Cl ₂	CH₂CH=CHCH₃
628	CH₃	4-OC ₆ H ₃ -2,4-Cl ₂	CH ₂ CH=C(CH ₃) ₂
629	CH₃	4-OC ₆ H ₃ -2,4-Cl ₂	CH₂CH=CHCI
630	CH ₃	4-OC ₆ H ₃ -2,4-Cl ₂	CH ₂ CH=CCl ₂
631	CH ₃	4-OC ₆ H ₃ -2,4-Cl ₂	CH ₂ C(CH ₃)=CH ₂
632	CH₃	4-OC ₆ H ₃ -2,4-Cl ₂	CH₂C≡CH
633	CH₃	4-OC ₆ H ₃ -2,4-Cl ₂	CH₂Si(CH₃)₃

Compound	R_2	(R ₅) _n	A-R ₇
No.		2	
634	CH ₃	4-OC ₆ H ₃ -2,4-Cl ₂	CH ₂ -c.propyl-2,2-Cl ₂
635	CH₃	4-OC ₆ H ₃ -2,4-Cl ₂	CH₂CN
636	CH₃	4-OC ₆ H ₃ -2,4-Cl ₂	CH ₂ COOC ₂ H ₅
637	CH₃	4-OC ₆ H ₃ -2,4-Cl ₂	CH(CH ₃)COOC ₂ H ₅
638	CH ₃	4-OC ₆ H ₃ -2,4-Cl ₂	CH ₂ C ₆ H ₄ -3-CF ₃
639	CH₃	4-OC ₆ H ₃ -2,4-Cl ₂	CH₂C ₆ H₄-4-F
640	CH₃	4-OC ₆ H ₃ -2,4-Cl ₂	CH ₂ C ₆ H ₄ -3-F
641	CH₃	4-OC ₆ H ₃ -2,4-Cl ₂	CH ₂ C ₆ H ₄ -2-F
642	CH₃	4-OC ₆ H ₃ -2,4-Cl ₂	C(=0)OC ₂ H ₅
643	CH₃	4-OC ₆ H ₃ -2,4-Cl ₂	C(=O)NHCH₃
644	CH₃	4-OC ₆ H ₃ -2,4-Cl ₂	C(=O)C(=O)OC ₂ H ₅
645	CH₃	4-OC ₆ H ₃ -3,4-Cl ₂	CH₃
646	CH₃	4-OC ₆ H ₃ -3,4-Cl ₂	C ₂ H ₅
647	CH ₃	4-OC ₆ H ₃ -3,4-Cl ₂	n-C ₃ H ₇
648	CH ₃	4-OC ₆ H ₃ -3,4-Cl ₂	i-C₃H ₇
649	CH₃	4-OC ₆ H ₃ -3,4-Cl ₂	n-C₄H ₉
650	CH₃	4-OC ₆ H ₃ -3,4-Cl ₂	n-C ₆ H ₁₃
651	CH₃	4-OC ₆ H ₃ -3,4-Cl ₂	CH₂F
652	CH ₃	4-OC ₆ H ₃ -3,4-Cl ₂	CHF₂
653	CH ₃	4-OC ₆ H ₃ -3,4-Cl ₂	CH₂CF₃
654	CH₃	4-OC ₆ H ₃ -3,4-Cl ₂	CH₂CH=CH₂
655	CH ₃	4-OC ₆ H ₃ -3,4-Cl ₂	CH₂CH=CHCH₃
656	CH₃	4-OC ₆ H ₃ -3,4-Cl ₂	CH₂CH=C(CH₃)₂
657	CH₃	4-OC ₆ H ₃ -3,4-Cl ₂	CH₂CH=CHCI
658	CH ₃	4-OC ₆ H ₃ -3,4-Cl ₂	CH2CH=CCl2
659	CH₃	4-OC ₆ H ₃ -3,4-Cl ₂	CH ₂ C(CH ₃)=CH ₂
660	CH₃	4-OC ₆ H ₃ -3,4-Cl ₂	CH₂C≡CH
661	CH₃	4-OC ₆ H ₃ -3,4-Cl ₂	CH₂Si(CH₃)₃
662	CH ₃	4-OC6H₃-3,4-Cl₂	CH ₂ -c.propyl-2,2-Cl ₂
663	CH ₃	4-OC ₆ H ₃ -3,4-Cl ₂	ĈH₂CN

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Compound	R ₂	(R ₅) _n	A-R ₇
No.		?	
664	CH₃	. 4-OC ₆ H ₃ -3,4-Cl₂	CH₂COOC₂H₅
665	CH₃	4-OC6H3-3,4-Cl2	CH(CH ₃)COOC ₂ H ₅
666	CH₃	4-OC ₆ H ₃ -3,4-Cl ₂	CH ₂ C ₆ H ₄ -3-CF ₃
667	CH₃	4-OC ₆ H ₃ -3,4-Cl ₂	CH ₂ C ₆ H ₄ -4-F
668	CH₃	4-OC ₆ H ₃ -3,4-Cl ₂	CH ₂ C ₆ H ₄ -3-F
669	CH₃	4-OC ₆ H ₃ -3,4-Cl ₂	CH₂C ₆ H₄-2-F
670	CH₃	4-OC ₆ H ₃ -3,4-Cl ₂	C(=O)OC ₂ H ₅
671	CH ₃	4-OC ₆ H ₃ -3,4-Cl ₂	C(=O)NHCH₃
672	CH ₃	4-OC ₆ H ₃ -3,4-Cl ₂	C(=O)C(=O)OC ₂ H ₅
673	CH₃	4-OC ₆ H ₃ -2-Cl,4-Br	CH₃
674	CH₃	4-OC ₆ H ₃ -2-Cl,4-Br	C ₂ H ₅
675	CH₃	4-OC ₆ H ₃ -2-Cl,4-Br	n-C₃H ₇
676	CH ₃	4-OC ₆ H ₃ -2-Cl,4-Br	i-C₃H ₇
677	CH₃	4-OC ₆ H ₃ -2-CI,4-Br	n-C₄H ₉
678	CH₃	4-OC ₆ H ₃ -2-Cl,4-Br	n-C ₆ H ₁₃
679	CH₃	4-OC ₆ H ₃ -2-Cl,4-Br	CH₂F
680	CH₃	4-OC ₆ H ₃ -2-Cl,4-Br	CHF ₂
681	CH ₃	4-OC ₆ H ₃ -2-Cl,4-Br	CH₂CF₃
682	CH₃	4-OC ₆ H ₃ -2-Cl,4-Br	CH₂CH=CH₂
683	CH₃	4-OC ₆ H ₃ -2-Cl,4-Br	CH₂CH=CHCH₃
684	CH₃	4-OC ₆ H ₃ -2-Cl,4-Br	CH₂CH=C(CH₃)₂
685	CH₃	4-OC ₆ H ₃ -2-Cl,4-Br	CH₂CH=CHCI
686	CH₃	4-OC ₆ H ₃ -2-Cl,4-Br	CH ₂ CH=CCl ₂
687	CH₃	4-OC ₆ H ₃ -2-Cl,4-Br	CH ₂ C(CH ₃)=CH ₂
688	CH₃	4-OC ₆ H ₃ -2-CI,4-Br	CH ₂ C≅CH
689	CH₃	4-OC ₆ H₃-2-Cl,4-Br	CH₂Si(CH₃)₃
690	CH₃	4-OC ₆ H ₃ -2-Cl,4-Br	CH ₂ -c.propyl-2,2-Cl ₂
691	CH₃	4-OC ₆ H ₃ -2-Cl,4-Br	CH₂CN
692	CH₃	4-OC ₆ H ₃ -2-Cl,4-Br	,,CH₂COOC₂H₅
693	CH₃	4-OC ₆ H ₃ -2-Cl,4-Br	CH(CH₃)COOC₂H₅

Compound	R ₂	(R ₅) _n	A-R ₇
No.	5	\$ - \$	
694	CH₃	4-OC ₆ H ₃ -2-Cl,4-Br	CH₂C ₆ H₄-3-CF₃
695	CH ₃	4-QC ₆ H ₃ -2-Cl,4-Br	C _{H2} C ₆ H ₄ -4-F
696	CH₃	4-OC ₆ H ₃ -2-Cl,4-Br	CH ₂ C ₆ H ₄ -3-F
697	CH₃	4-OC ₆ H ₃ -2-Cl,4-Br	CH ₂ C ₆ H ₄ -2-F
698	CH₃	4-OC ₆ H ₃ -2-Cl,4-Br	C(=O)OC ₂ H ₅
699	CH₃	4-OC ₆ H ₃ -2-Cl,4-Br	C(=O)NHCH ₃
700	CH₃	4-OC ₆ H ₃ -2-Cl,4-Br	$C(=O)C(=O)OC_2H_5$
701	CH₃	4-OC ₆ H ₃ -3,4-(-OCH ₂ O-)	CH₃
702	CH₃	4-OC ₆ H ₃ -3,4-(-OCH ₂ O-)	C ₂ H ₅
703	CH ₃	4-OC ₆ H ₃ -3,4-(-OCH ₂ O-)	n-C₃H ₇
704	CH₃	4-OC ₆ H ₃ -3,4-(-OCH ₂ O-)	i-C₃H ₇
705	CH₃	4-OC ₆ H ₃ -3,4-(-OCH ₂ O-)	n-C₄H ₉
706	CH ₃	4-OC ₆ H ₃ -3,4-(-OCH ₂ O-)	n-C ₆ H ₁₃
707	CH₃	4-OC ₆ H ₃ -3,4-(-OCH ₂ O-)	CH₂F
708	CH ₃	4-OC ₆ H ₃ -3,4-(-OCH ₂ O-)	CHF₂
709	CH ₃	4-OC ₆ H ₃ -3,4-(-OCH ₂ O-)	CH₂CF₃
710	CH₃	4-OC ₆ H ₃ -3,4-(-OCH ₂ O-)	CH₂CH=CH₂
711	CH₃	4-OC ₆ H ₃ -3,4-(-OCH ₂ O-)	CH₂CH=CHCH₃
712	CH ₃	4-OC ₆ H ₃ -3,4-(-OCH ₂ O-)	CH ₂ CH=C(CH ₃) ₂
713	CH ₃	4-OC ₆ H ₃ -3,4-(-OCH ₂ O-)	CH₂CH=CHCI
714	CH ₃	4-OC ₆ H ₃ -3,4-(-OCH ₂ O-)	CH₂CH=CCI₂
715	CH₃	4-OC ₆ H ₃ -3,4-(-OCH ₂ O-)	CH ₂ C(CH ₃)=CH ₂
716	CH ₃	4-OC ₆ H ₃ -3,4-(-OCH ₂ O-)	CH ₂ C≡CH
717	CH₃	4-OC ₆ H ₃ -3,4-(-OCH ₂ O-)	CH₂Si(CH₃)₃
718	CH₃	4-OC ₆ H ₃ -3,4-(-OCH ₂ O-)	CH ₂ -c.propyl-2,2-Cl ₂
719	CH₃	4-OC ₆ H ₃ -3,4-(-OCH ₂ O-)	CH₂CN
720	CH₃	4-OC ₆ H ₃ -3,4-(-OCH ₂ O-)	CH ₂ COOC ₂ H ₅
721	CH₃	4-OC ₆ H ₃ -3,4-(-OCH ₂ O-)	CH(CH₃)COOC₂H₅
722	CH ₃	4-OC ₆ H ₃ -3,4-(-OCH ₂ O-)	CH₂C ₆ H₄-3-CF₃
723	CH ₃	4-OC ₆ H ₃ -3,4-(-OCH ₂ O-)	CH ₂ C ₆ H ₄ -4-F

Compound	R ₂	(R ₅) _n	A-R ₇
No.	3	,	
724	CH₃	4-OC ₆ H₃-3,4-(-OCH₂O-)	CH₂C ₆ H₄-3-F
725	CH₃	4-OC ₆ H ₃ -3,4-(-OCH ₂ O-)	CH₂C ₆ H₄-2-F
726	CH₃	4-OC ₆ H ₃ -3,4-(-OCH ₂ O-)	C(=O)OC ₂ H ₅
727	CH₃	4-OC ₆ H ₃ -3,4-(-OCH ₂ O-)	C(=O)NHCH₃
728	CH₃	4-OC ₆ H ₃ -3,4-(-OCH ₂ O-)	$C(=O)C(=O)OC_2H_5$
729	CH ₃	4-O _, C₀H₄-4-SCH₃	CH ₃
730	CH ₃	4-OC ₆ H ₄ -4-SCH ₃	C ₂ H ₅
731	CH ₃	4-OC ₆ H ₄ -4-SCH ₃	n-C ₃ H ₇
732	CH ₃	4-OC ₆ H ₄ -4-SCH ₃	i-C ₃ H ₇
733	CH3	4-OC ₆ H₄-4-SCH₃	n-C₄H ₉
734	CH₃	4-OC ₆ H ₄ -4-SCH ₃	n-C ₆ H ₁₃
735	CH₃	4-OC ₆ H₄-4-SCH₃	CH₂F
736	CH₃	4-OC ₆ H₄-4-SCH₃	CHF ₂
737	CH₃	4-OC ₆ H₄-4-SCH₃	CH₂CF₃
738	CH₃	4-OC ₆ H ₄ -4-SCH ₃	CH ₂ CH=CH ₂
739	CH₃	4-OC ₆ H ₄ -4-SCH ₃	CH₂CH=CHCH₃
740	CH ₃	4-OC ₆ H ₄ -4-SCH ₃	CH ₂ CH=C(CH ₃) ₂
741	CH₃	4-OC ₆ H ₄ -4-SCH ₃	CH₂CH=CHCI
742	CH₃	4-OC ₆ H ₄ -4-SCH ₃	CH₂CH=CCI₂
743	CH₃	4-OC ₆ H ₄ -4-SCH ₃	CH ₂ C(CH ₃)=CH ₂
744	CH ₃	4-OC ₆ H ₄ -4-SCH ₃	CH₂C≅CH
745	CH₃	4-OC ₆ H ₄ -4-SCH ₃	CH₂Si(CH₃)₃
746	CH₃	4-OC,6H4-4-SCH3	CH ₂ -c.propyl-2,2-Cl ₂
747	CH₃	4-OC ₆ H ₄ -4-SCH ₃	CH ₂ CN
748	CH₃	4-OC ₆ H ₄ -4-SCH ₃	CH2COOC2H5
749	CH₃	4-OC ₆ H ₄ -4-SCH ₃	CH(CH ₃)COOC ₂ H ₅
750	CH ₃	4-OC ₆ H ₄ -4-SCH ₃	CH₂C₅H₄-3-CF₃
751	CH₃	4-OC ₆ H ₄ -4-SCH ₃	CH₂C ₆ H₄-4-F
752	CH ₃	4-OC ₆ H ₄ -4-SCH ₃	CH₂C ₆ H₄-3-F
753	CH₃	4-OC ₆ H ₄ -4-SCH ₃	ີັCH₂C₅H₄-2-F

Compound	R ₂	(R ₅) _n	A-R ₇
No.		·	
754	CH₃	4-OC ₆ H₄-4-SCH₃	C(=O)OC ₂ H ₅
755	CH₃	4-OC ₆ H₄-4-SCH₃	C(=O)NHCH ₃
756	CH₃	4-OC₅H ₄ -4-SCH₃	C(=0)C(=0)OC ₂ H ₅
757	CH₃	4-OC ₆ H ₄ -4-OCH ₃	CH₃
758	CH₃	4-OC ₆ H ₄ -4-OCH ₃	C ₂ H ₅
759	CH₃	4-OC ₆ H ₄ -4-OCH ₃	n-C ₃ H ₇
760	CH ₃	4-OC ₆ H ₄ -4-OCH ₃	i-C ₃ H ₇
761	CH₃	4-OC ₆ H ₄ -4-OCH ₃	n-C₄H ₉
762	CH₃	4-OC ₆ H ₄ -4-OCH ₃	n-C ₆ H ₁₃
763	CH₃	4-OC ₆ H ₄ -4-OCH ₃	CH₂F
764	CH₃	4-OC ₆ H₄-4-OCH₃	CHF ₂
765	CH ₃	4-OC ₆ H ₄ -4-OCH ₃	CH ₂ CF ₃
766	CH₃	4-OC ₆ H ₄ -4-OCH ₃	CH₂CH=CH₂
767	CH₃	4-OC ₆ H ₄ -4-OCH ₃	CH₂CH=CHCH₃
768	CH ₃	4-OC ₆ H ₄ -4-OCH ₃	CH ₂ CH=C(CH ₃) ₂
769	CH₃	4-OC ₆ H ₄ -4-OCH ₃	CH₂CH=CHCI
770	CH₃	4-OC ₆ H ₄ -4-OCH ₃	CH ₂ CH=CCI ₂
771	CH₃	4-OC ₆ H ₄ -4-OCH ₃	CH ₂ C(CH ₃)=CH ₂
772	CH ₃	4-OC ₆ H ₄ -4-OCH ₃	CH₂C≊CH
773	CH ₃	4-OC ₆ H ₄ -4-OCH ₃	CH ₂ Si(CH ₃) ₃
774	CH₃	4-OC ₆ H ₄ -4-OCH ₃	CH ₂ -c.propyl-2,2-Cl ₂
775	CH₃	4-OC ₆ H ₄ -4-OCH ₃	CH₂CN
776	CH₃	4-OC ₆ H ₄ -4-OCH ₃	CH₂COOC₂H₅
777	CH₃	4-OC ₆ H ₄ -4-OCH ₃	CH(CH₃)COOC₂H₅
778	CH ₃	4-OC ₆ H ₄ -4-OCH ₃	CH ₂ C ₆ H ₄ -3-CF ₃
779	CH₃	4-OC ₆ H ₄ -4-OCH ₃	CH₂C ₆ H₄-4-F
780	CH₃	4-OC ₆ H ₄ -4-OCH ₃	CH ₂ C ₆ H ₄ -3-F
781	CH₃	4-OC ₆ H ₄ -4-OCH ₃	CH₂C ₆ H₄-2-F
782	CH₃	4-OC ₆ H ₄ -4-OCH ₃	_C(=O)OC ₂ H ₅
783	CH ₃	4-OC ₆ H ₄ -4-OCH ₃	C(=O)NHCH₃

Compound	i R ₂	(R ₅) _n	A-R ₇
No.		3	
784	CH₃	4-OC ₆ H ₄ -4-OCH ₃	C(=O)C(=O)OC ₂ H ₅
785	CH₃	4-OC ₆ H ₄ -4-t-butyl	CH ₃
786	CH₃	4-OC ₆ H ₄ -4-t-butyl	C₂H₅
787	CH₃	4-OC ₆ H ₄ -4-t-butyl	n-C ₃ H ₇
788	CH₃	4-OC ₆ H ₄ -4-t-butyl	i-C₃H ₇
789	CH ₃	4-OC ₆ H₄-4-t-butyl	n-C ₄ H ₉
790	CH₃	4-OC ₆ H₄-4-t-butyl	n-C ₆ H ₁₉
791	CH ₃	4-OC ₆ H ₄ -4-t-butyl	CH₂F
792	CH₃	4-OC ₆ H ₄ -4-t-butyl	CHF₂
793	CH₃	4-OC ₆ H ₄ -4-t-butyl	CH₂CF₃
794	CH₃	4-OC ₆ H ₄ -4-t-butyl	CH ₂ CH=CH ₂
795	CH ₃	4-OC ₆ H ₄ -4-t-butyl	CH₂CH=CHCH₃
796	CH₃	4-OC ₆ H ₄ -4-t-butyl	CH₂CH≃C(CH₃)₂
797	CH ₃	4-OC ₆ H ₄ -4-t-butyl	CH₂CH=CHCI
798	CH ₃	4-OC ₆ H ₄ -4-t-butyl	CH ₂ CH=CCl ₂
799	CH₃	4-OC ₆ H ₄ -4-t-butyl	CH ₂ C(CH ₃)=CH ₂
800	CH₃	4-OC ₆ H ₄ -4-t-butyl	CH ₂ C≡CH
801	СН₃	4-OC ₆ H₄-4-t-butyl	CH₂Sì(CH₃)₃
802	CH₃	4-OC ₆ H₄-4-t-butyl	CH ₂ -c.propyl-2,2-Cl ₂
803	CH₃	4-OC ₆ H₄-4-t-butyl	CH₂CN
804	CH₃	4-OC ₆ H ₄ -4-t-butyl	CH₂COOC₂H₅
805	CH ₃	4-OC ₆ H ₄ -4-t-butyl	CH(CH₃)COOC₂H₅
806	CH₃	4-OC ₆ H ₄ -4-t-butyl	CH ₂ C ₆ H ₄ -3-CF ₃
807	CH₃	4-OC ₆ H ₄ -4-t-butyl	CH ₂ C ₆ H ₄ -4-F
808	CH₃	4-OC ₆ H ₄ -4-t-butyl	CH ₂ C ₆ H₄-3-F
809	CH₃	4-OC ₆ H ₄ -4-t-butyl	CH₂C ₆ H₄-2-F
810	CH₃	4-OC ₆ H₄-4-t-butyl	$C(=O)OC_2H_5$
811	CH ₃	4-OC ₆ H₄-4-t-butyl	C(=O)NHCH₃
812	CH₃	4-OC ₆ H ₄ -4-t-butyl	©(=0)C(=0)OC ₂ H ₅
813	CH ₃	4-OC ₆ H ₄ -4-CF ₃	CH₃

Compound	R ₂	(R₅) _n	A-R ₇
No.		3 ,	
814	CH₃	4-OC ₆ H ₄ -4-CF ₃	C₂H₅
815	CH₃	4-OC ₆ H ₄ -4-CF ₃	n-C₃H ₇
816	CH₃	4-OC ₆ H ₄ -4-CF ₃	i-C₃H ₇
817	CH₃	4-OC ₆ H ₄ -4-CF ₃	n-C₄H ₉
818	CH ₃	4-OC ₆ H ₄ -4-CF ₃	n-C ₆ H ₁₃
819	CH ₃	4-OC ₆ H ₄ -4-CF ₃	CH₂F
820	CH₃	4-OC ₆ H ₄ -4-CF ₃	CHF ₂
821	CH₃	4-OC6H4-4-CF3	CH₂CF₃
822	CH₃	4-OC ₆ H ₄ -4-CF ₃	CH₂CH=CH₂
823	CH ₃	4-OC ₆ H ₄ -4-CF ₃	CH₂CH=CHCH₃
824	CH₃	4-OC ₆ H ₄ -4-CF ₃	CH ₂ CH=C(CH ₃) ₂
825	CH₃	4-OC ₆ H ₄ -4-CF ₃	CH₂CH=CHCI
826	CH₃	4-OC ₆ H ₄ -4-CF ₃	CH ₂ CH=CCI ₂
827	CH₃	4-OC ₆ H ₄ -4-CF ₃	CH ₂ C(CH ₃)=CH ₂
828 ·	CH ₃	4-OC ₆ H ₄ -4-CF ₃	CH₂C≡CH
829	CH ₃	4-OC ₆ H ₄ -4-CF ₃	CH₂Si(CH₃)₃
830	CH₃	4-OC ₆ H ₄ -4-CF ₃	CH ₂ -c.propyl-2,2-Cl ₂
831	CH₃	4-OC ₆ H ₄ -4-CF ₃	CH₂CN
832	CH ₃	4-OC ₆ H ₄ -4-CF ₃	CH₂COOC₂H₅
833	CH₃	4-OC ₆ H ₄ -4-CF ₃	CH(CH₃)COOC₂H₅
834	CH ₃	4-OC ₆ H ₄ -4-CF ₃	CH ₂ C ₆ H ₄ -3-CF ₃
835	CH₃	4-OC ₆ H ₄ -4-CF ₃	CH ₂ C ₆ H ₄ -4-F
836	CH₃	4-OC ₆ H ₄ -4-CF ₃	CH ₂ C ₆ H ₄ -3-F
837	CH ₃	4-OC ₆ H ₄ -4-CF ₃	CH₂C ₆ H₄-2-F
838	CH₃	4-OC ₆ H ₄ -4-CF ₃	$C(=O)OC_2H_5$
839	CH₃	4-OC ₆ H ₄ -4-CF ₃	C(=O)NHCH ₃
840	CH₃	4-OC ₆ H ₄ -4-CF ₃	$C(=0)C(=0)OC_2H_5$
841	CH₃	4-OC ₆ H ₄ -2-CF ₃	CH ₃
842	CH₃	4-OC ₆ H ₄ -2-CF ₃	C ₂ H ₅
843	CH₃	4-OC ₆ H ₄ -2-CF ₃	n-C ₃ H ₇

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3
3
I=CH ₂
l=CHCH₃
=C(CH ₃) ₂
=CHCI
=CCI ₂
CH ₃)=CH ₂
СН
CH₃)₃
ropyl-2,2-Cl₂
OC₂H₅
₃)COOC₂H₅
I₄-3-CF₃
1 ₄ -4-F
I₄-3-F
I₄-2-F
C₂H₅
IHCH₃
(=0)0C ₂ H ₅

Compound	R ₂	(R ₅) _n	A-R ₇
No.		1	
874	CH₃	4-OCH₂C ₆ H₄-4-Cl	n-C ₆ H ₁₃
875	CH₃	4-OCH ₂ C ₆ H ₄ -4-CI	CH₂F
876	CH ₃	4-OCH ₂ C ₆ H ₄ -4-CI	CHF ₂
877	CH ₃	4-OCH ₂ C ₆ H ₄ -4-CI	CH₂CF₃
878	CH₃	4-OCH ₂ C ₆ H ₄ -4-Cl	CH₂CH=CH₂
879	CH ₃	4-0CH ₂ C ₆ H ₄ -4-Cl	CH₂CH=CHCH₃
088	CH₃	4-OCH ₂ C ₆ H ₄ -4-Cl	CH ₂ CH=C(CH ₃) ₂
881	CH₃	4-OCH ₂ C ₆ H ₄ -4-Cl	CH₂CH=CHCI
882	CH₃	4-OCH ₂ C ₆ H ₄ -4-Cl	CH ₂ CH=CCl ₂
. 883	CH₃	4-OCH ₂ C ₆ H ₄ -4-Cl	CH ₂ C(CH ₃)=CH ₂
884	CH₃	4-OCH ₂ C ₆ H ₄ -4-Cl	CH₂C≡CH
885	CH₃	4-OCH ₂ C ₆ H ₄ -4-Cl	CH₂Si(CH₃)₃
886	CH₃	4-OCH ₂ C ₆ H ₄ -4-Cl	CH ₂ -c.propyl-2,2-Cl ₂
887	CH₃	4-OCH ₂ C ₅ H ₄ -4-Cl	CH₂CN
888	CH ₃	4-OCH ₂ C ₆ H ₄ -4-Cl	CH ₂ COOC ₂ H ₅
889	CH ₃	4-OCH ₂ C ₆ H ₄ -4-Cl	CH(CH₃)COOC₂H₅
890	CH₃	4-OCH ₂ C ₆ H ₄ -4-Cl	CH₂C ₆ H₄-3-CF₃
891	CH₃	4-OCH ₂ C ₆ H ₄ -4-Cl	CH₂C ₆ H₄-4-F
892	CH ₃	4-OCH ₂ C ₆ H ₄ -4-Cl	CH₂C ₆ H₄-3-F
893	CH₃	4-OCH ₂ C ₆ H ₄ -4-Cl	CH ₂ C ₆ H ₄ -2-F
894	CH₃	4-OCH ₂ C ₆ H ₄ -4-Cl	$C(=O)OC_2H_5$
895	CH₃	4-OCH ₂ C ₆ H ₄ -4-Cl	C(=O)NHCH ₃
896	CH ₃	4-OCH ₂ C ₆ H ₄ -4-Cl	$C(=O)C(=O)OC_2H_5$
797	CH3	4-OCH ₂ C ₆ H ₃ -3,4-Cl ₂	CH₃
898	CH₃	4-OCH ₂ C ₆ H ₃ -3,4-Cl ₂	C₂H₅
899	CH ₃	4-OCH ₂ C ₆ H ₃ -3,4-Cl ₂	n-C₃H ₇
900	CH ₃	4-OCH ₂ C ₆ H ₃ -3,4-Cl ₂	i-C ₃ H ₇
901	CH₃	4-OCH ₂ C ₆ H ₃ -3,4-Cl ₂	n-C₄H ₉
902	CH ₃	4-OCH ₂ C ₆ H ₃ -3,4-Cl ₂	α-C ₆ H ₁₃
903	CH ₃	4-OCH ₂ C ₆ H ₃ -3,4-Cl ₂	CH₂F

Compound No.	R ₂	(R ₅) _n	A-R ₇
904	CH ₃	4-OCH ₂ C ₆ H ₃ -3,4-Cl ₂	CHF ₂
905	CH ₃	4-OCH ₂ C ₆ H ₃ -3,4-Cl ₂	CH ₂ CF ₃
906	CH₃	4-OCH ₂ C ₆ H ₃ -3,4-Cl ₂	CH ₂ CH=CH ₂
907	CH₃	4-OCH2C6H3-3,4-Cl2	CH₂CH=CHCH₃
908	CH₃	4-OCH ₂ C ₆ H ₃ -3,4-Cl ₂	CH ₂ CH=C(CH ₃) ₂
909	CH₃	4-OCH ₂ C ₆ H ₃ -3,4-Cl ₂	CH₂CH=CHCI
910	CH₃	4-OCH ₂ C ₆ H ₃ -3,4-Cl ₂	CH ₂ CH=CCl ₂
911	СН₃	4-OCH ₂ C ₆ H ₃ -3,4-Cl ₂	CH₂C(CH₃)=CH₂
912	CH₃	4-OCH ₂ C ₆ H ₃ -3,4-Cl ₂	CH₂C≡CH
913	СН₃	4-OCH ₂ C ₆ H ₃ -3,4-Cl ₂	CH ₂ Si(CH ₃) ₃
914	CH₃	4-OCH ₂ C ₆ H ₃ -3,4-Cl ₂	CH ₂ -c.propyl-2,2-Cl ₂
915	CH₃	4-OCH ₂ C ₆ H ₃ -3,4-Cl ₂	CH₂CN
916	CH ₃	4-OCH ₂ C ₆ H ₃ -3,4-Cl ₂	CH₂COOC₂H₅
917	CH₃	4-OCH ₂ C ₆ H ₃ -3,4-Cl ₂	CH(CH₃)COOC₂H₅
918	CH₃	4-OCH ₂ C ₆ H ₃ -3,4-Cl ₂	CH₂C ₆ H₄-3-CF₃
919	CH₃	4-OCH ₂ C ₆ H ₃ -3,4-Cl ₂	CH₂C ₆ H₄-4-F
920	CH₃	4-OCH ₂ C ₆ H ₃ -3,4-Cl ₂	CH ₂ C ₆ H ₄ -3-F
921	CH₃	4-OCH ₂ C ₆ H ₃ -3,4-Cl ₂	CH ₂ C ₆ H ₄ -2-F
922	CH ₃	4-OCH ₂ C ₆ H ₃ -3,4-Cl ₂	C(=O)OC ₂ H ₅
923	CH₃	4-OCH ₂ C ₆ H ₃ -3,4-Cl ₂	C(=O)NHCH ₃
924	CH₃	4-OCH ₂ C ₆ H ₃ -3,4-Cl ₂	C(=O)C(=O)OC ₂ H ₅

In Tables 2.1 and 2.2, the 13 C-NMR data of the compounds 1-[4-(3-trifluoromethyl-phenylmethoxy)-phenyl]-1,2-propanedione 1-E-[methyloxime]-2-oxime and 1-[4-(3-trifluoromethylphenylmethoxy)-phenyl]-1,2-propanedione 1-Z-[methyloxime]-2-oxime (which was prepared by one of the known processes and from which the E/Z isomer mixture formed in the preparation was isolated) or, respectively, methyl 2-[[[(1-methyl-2-(4-(3-trifluoromethylphenylmethoxy)-phenyl)-2-E-[methoxyimino]ethylidene)amino]oxy]methyl]- α -(methoxymethylene)-phenylacetate (compound A225 in Table 1) are shown. The similar

chemical shifts of atoms 1 and 4 of compound A in Table 2.1 and those in Table 2.2 confirm the E configuration of the compounds of the formula I.

<u>Table 2.1</u>: ¹³C-NMR shifts and ¹J_{CC} coupling constants of 1-[4-(3-trifluoromethylphenylmethoxy)-phenyl]-1,2-propanedione 1-E-[methyloxime]-2-oxime (A) and 1-[4-(3-trifluoromethylphenylmethoxy)-phenyl]-1,2-propanedione 1-Z-[methyloxime]-2-oxime (B)

Compound	Atom No.	Shift δ (ppm)	Coupling ¹ Jcc (Hz)
Α	1	125.6	$J_{12} = 56.0$
	3	155.0	$J_{23} = 72.0$
	4 .	10.1	$J_{34} = 43.0$
В	1 '	127.8	$J_{12} = 69.0$
	3	152.1	$J_{23} = 56.5$
	4	14.4	$J_{34} = 41.5$

<u>Table 2.2</u>: 13 C-NMR shifts of methyl 2-[[[(1-methyl-2-(4-(3-trifluoromethylphenylmethoxy)-phenyl)-2-E-[methoxyimino]ethylidene)amino]oxy]methyl]- α -(methoxymethylene)-phenylacetate (compound 1.225)

Atom No.	Shift δ (ppm)
1	124.9
2	155.1
3	155.0
4	11.1